

# What is spectrum?

**Continuum** + **Spectral Features**

**Blackbody**

free-free

- 1 blackbody**
- 2 radiation**
- 3 Thomson scattering**
- 4 radiation transfer**
- 5 Einstein coefficients**

**absorption / emission**

atom  
molecules

**high res.**

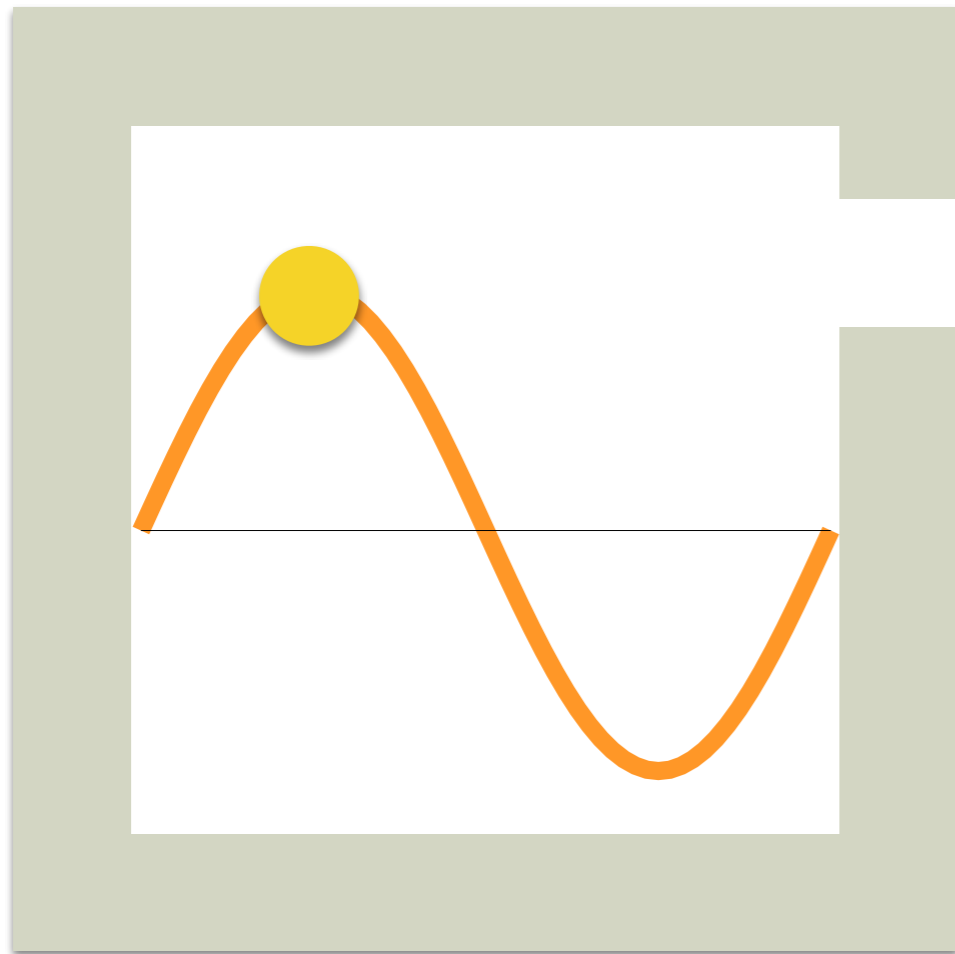
ice  
dust

**low res.**

# Blackbody from scratch

The probability that a system is in the energy level  $\epsilon_s$  is proportional to

$$p \propto \exp\left(-\frac{\epsilon_s}{\tau}\right)$$



**system**

**mode**

**energy**

**temperature**

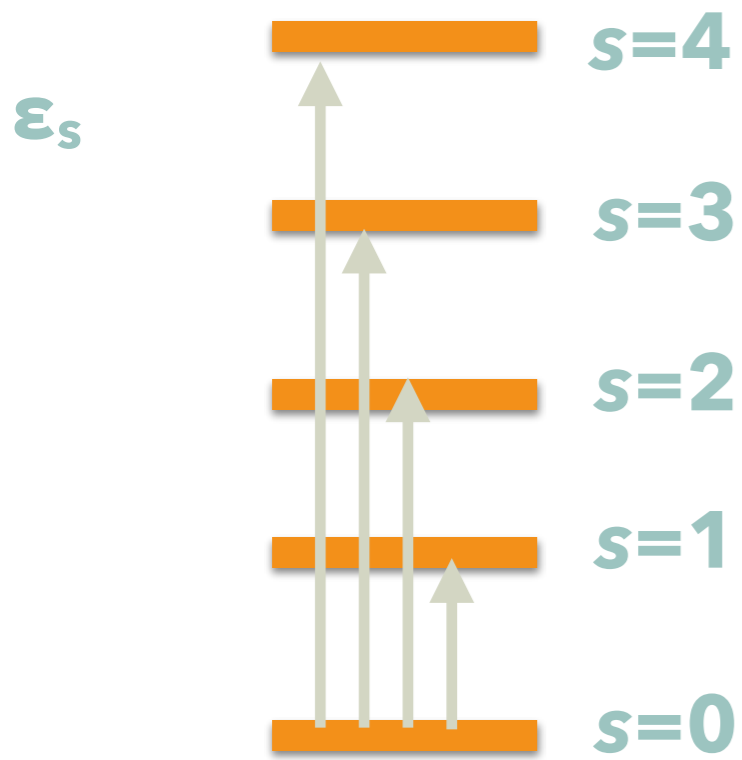
**photons in a cavity**

$\omega$  ( $=2\pi\nu$ )

$\hbar\omega$

$\tau$  ( $=kT$ )

- 1** how many photons are in the mode  $\omega$ ?
- 2** how many modes are in whole space?  
(phase space)

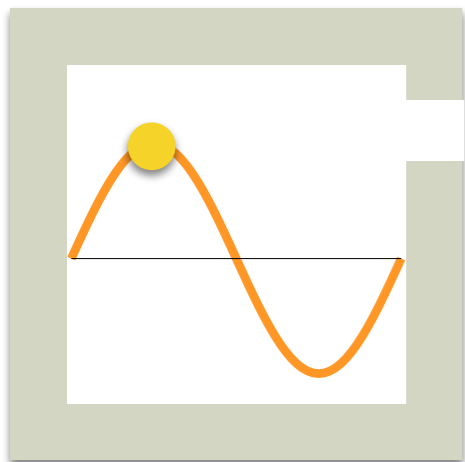
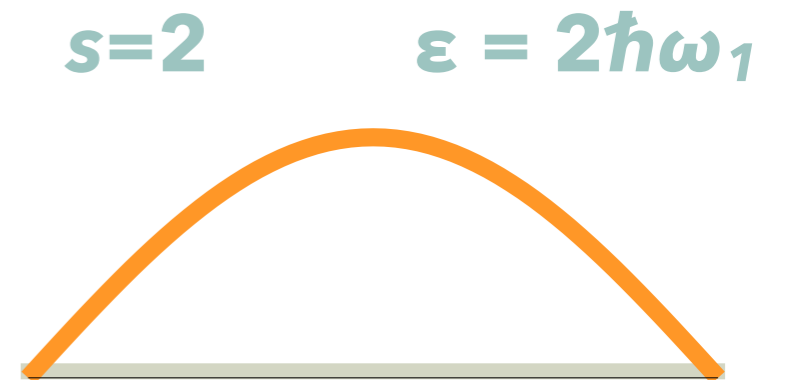
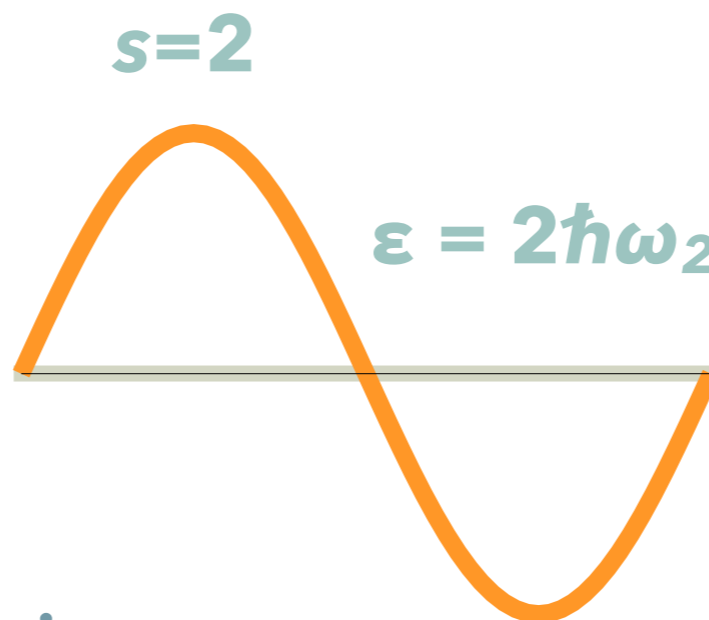
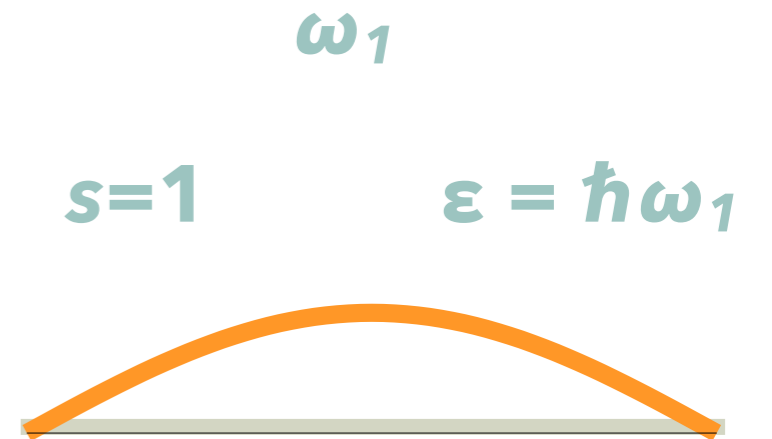
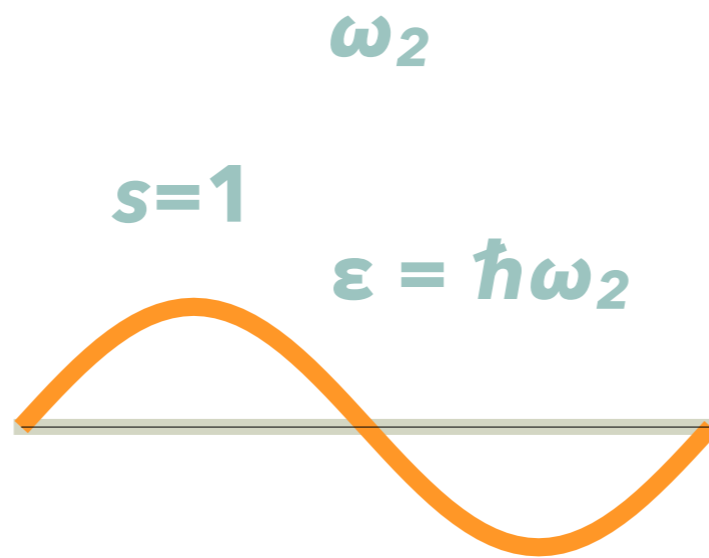


$$\epsilon_s = s \hbar \omega_1$$

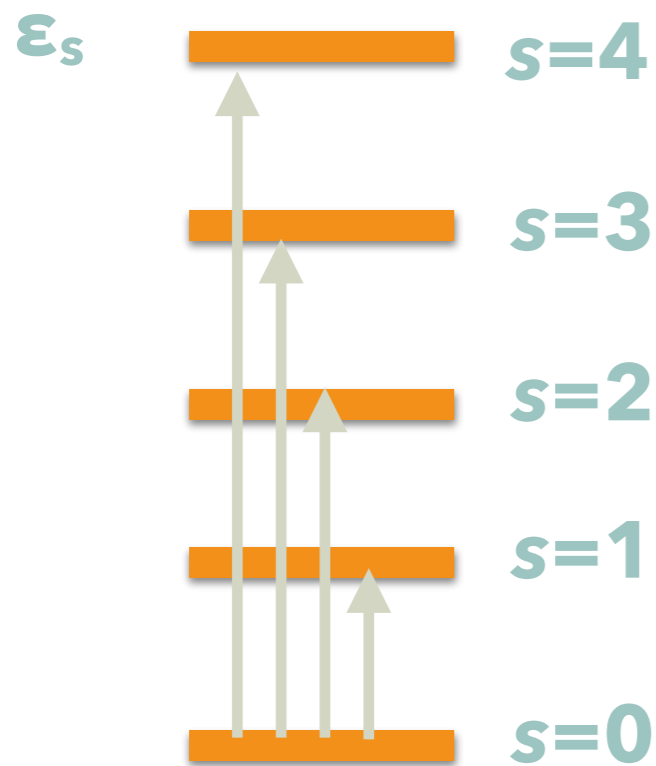
$$p \propto \exp\left(-\frac{\epsilon_s}{\tau}\right)$$

**talking about a single mode currently**

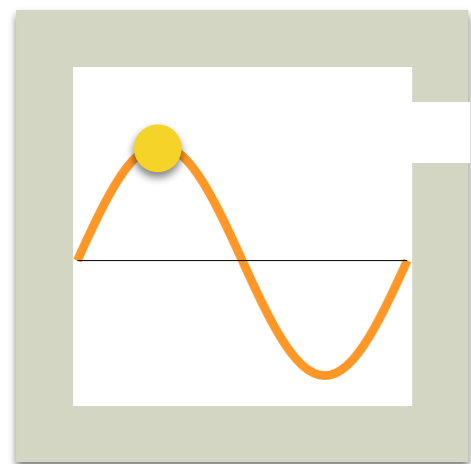
**$s$  photon in  $\omega$  is equivalent to an oscillator being in  $s$ -th orbital**



**how many photons are in the mode  $\omega$ ?**



$s$  photon in  $\omega$  is equivalent to an oscillator being in  $s$ -th orbital



**1**

how many photons are in the mode  $\omega$ ?

partition function

(sum of all possibilities)

$$Z = \sum_s \exp\left(-\frac{\epsilon_s}{\tau}\right)$$

$$= \sum_s \exp\left(-\frac{s\hbar\omega}{\tau}\right)$$

this is a geometric progression

$$Z = \frac{1}{1 - \exp\left(-\frac{\hbar\omega}{\tau}\right)}$$

average occupation of  $\omega$

$$\langle s \rangle = \frac{\sum_s s \exp\left(-\frac{s\hbar\omega}{\tau}\right)}{Z}$$

$$p \propto \exp\left(-\frac{\epsilon_s}{\tau}\right)$$

$$\epsilon_s = s\hbar\omega$$

$$\sum_{i=1}^n x^{i-1} = \frac{1-x^n}{1-x}$$

$$x < 1$$

$$\frac{1}{1-x}$$

photons

## average occupation of $\omega$

$$\langle s \rangle = \frac{\sum_s s \exp(-\frac{s\hbar\omega}{\tau})}{Z}$$

$$Z = \frac{1}{1 - \exp(-\frac{\hbar\omega}{\tau})}$$
$$y = \frac{\hbar\omega}{\tau}$$

$$\sum_s s \exp(-sy) = -\frac{d}{dy} \sum_s \exp(-sy)$$

$$= \frac{d}{dy} \frac{-1}{1 - \exp(-y)} = \frac{\exp(-y)}{[1 - \exp(-y)]^2}$$

$$\langle s \rangle = \frac{\exp(-y)}{1 - \exp(-y)} = \frac{1}{\exp(y) - 1}$$

$$\sum_{i=1}^n x^{i-1} = \frac{1-x^n}{1-x}$$

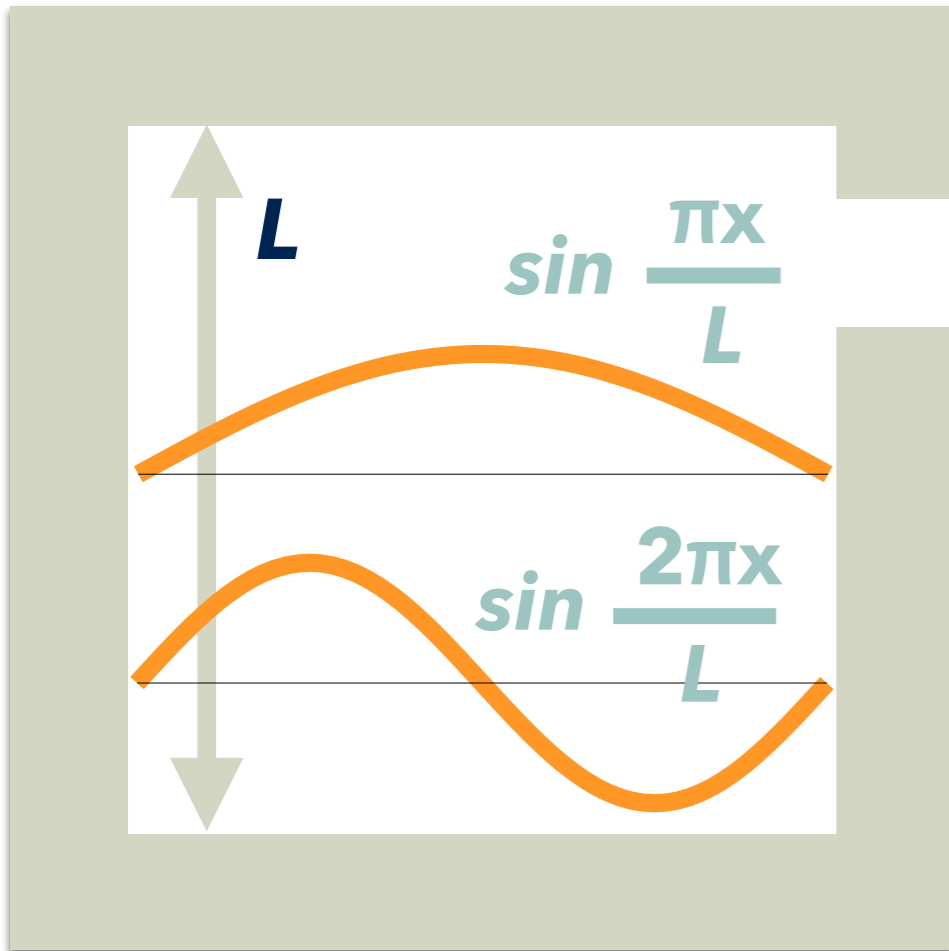
$$\langle \varepsilon \rangle = \frac{\hbar\omega}{\exp(\frac{\hbar\omega}{\tau}) - 1}$$

$$x < 1$$

$$\frac{1}{1-x}$$

average energy in mode  $\omega$

## standing wave



## wave equation

$$\frac{\partial^2 E}{\partial t^2} = c^2 \frac{\partial^2 E}{\partial x^2}$$

$$\mathbf{n} = (n_x, n_y, n_z)$$

propagation of wave

## 2 how many modes are in phase space

$$\mathbf{E} = (E_x, E_y, E_z)$$

$$E_x = E_{x0} \sin \omega t \cos \frac{n_x \pi x}{L} \sin \frac{n_y \pi y}{L} \sin \frac{n_z \pi z}{L}$$

**temporal**

$$E_y = E_{y0} \sin \omega t \sin \frac{n_x \pi x}{L} \cos \frac{n_y \pi y}{L} \sin \frac{n_z \pi z}{L}$$

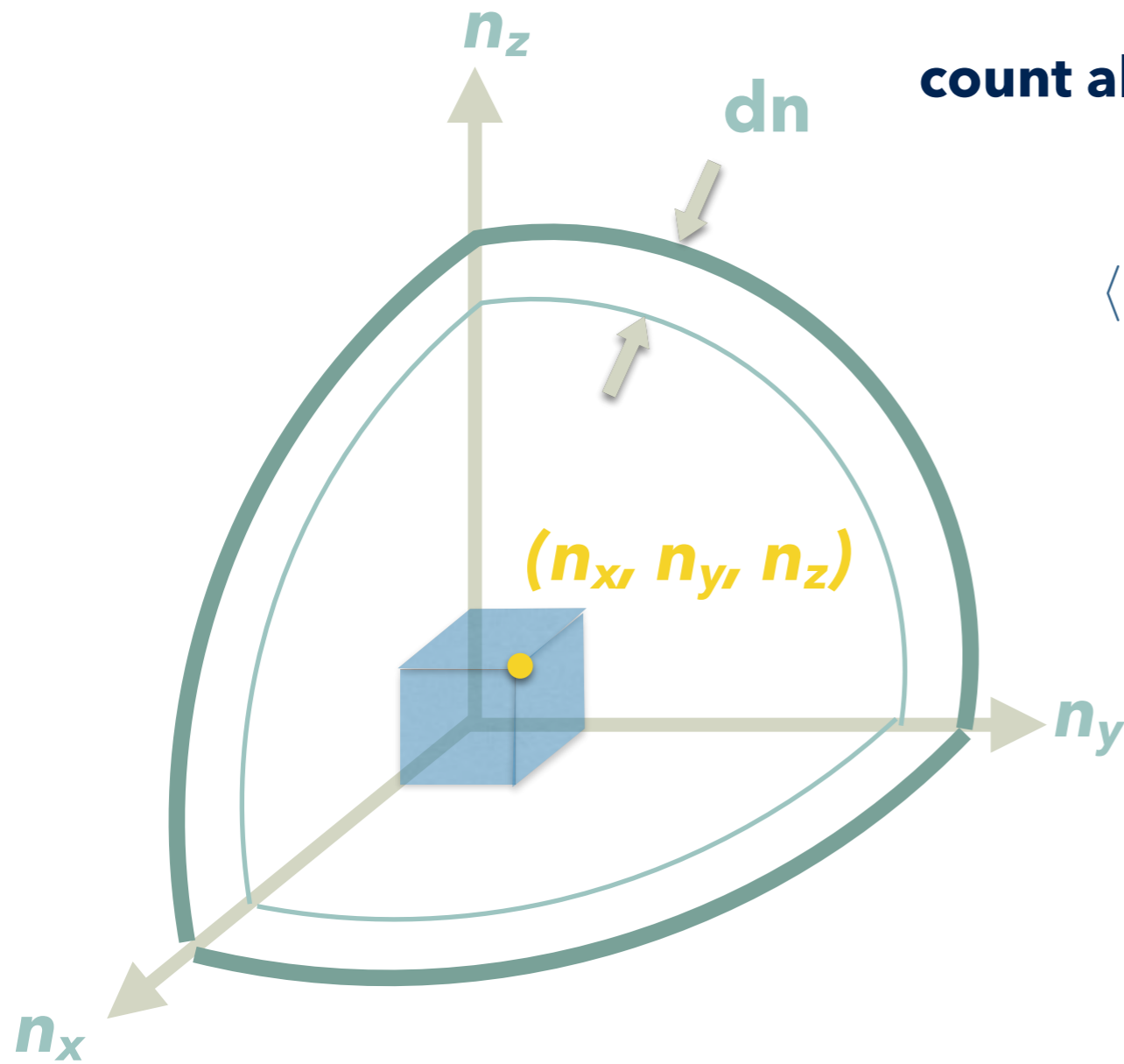
$$E_z = E_{z0} \sin \omega t \sin \frac{n_x \pi x}{L} \sin \frac{n_y \pi y}{L} \cos \frac{n_z \pi z}{L}$$

$$\omega^2 = c^2 \left[ \left( \frac{n_x \pi}{L} \right)^2 + \left( \frac{n_y \pi}{L} \right)^2 + \left( \frac{n_z \pi}{L} \right)^2 \right]$$

$$n^2 = n_x^2 + n_y^2 + n_z^2$$

$$\omega = \frac{n \pi c}{L} \quad \text{mode}$$

we already know how the mode is occupied



count all modes in phase space

$$\omega = \frac{n\pi c}{L}$$

$$n = \frac{\omega L}{\pi c}$$

$$\langle \epsilon \rangle = \frac{\hbar\omega}{\exp \frac{\hbar\omega}{\tau} - 1}$$

$$U = 4\pi n^2 \int_0^\infty \langle \epsilon \rangle dn \times \frac{1}{8} \times 2$$

positive polarization

$$U = 4\pi n^2 \int_0^\infty \frac{\hbar\omega}{\exp \frac{\hbar\omega}{\tau} - 1} dn \times \frac{1}{8} \times 2$$

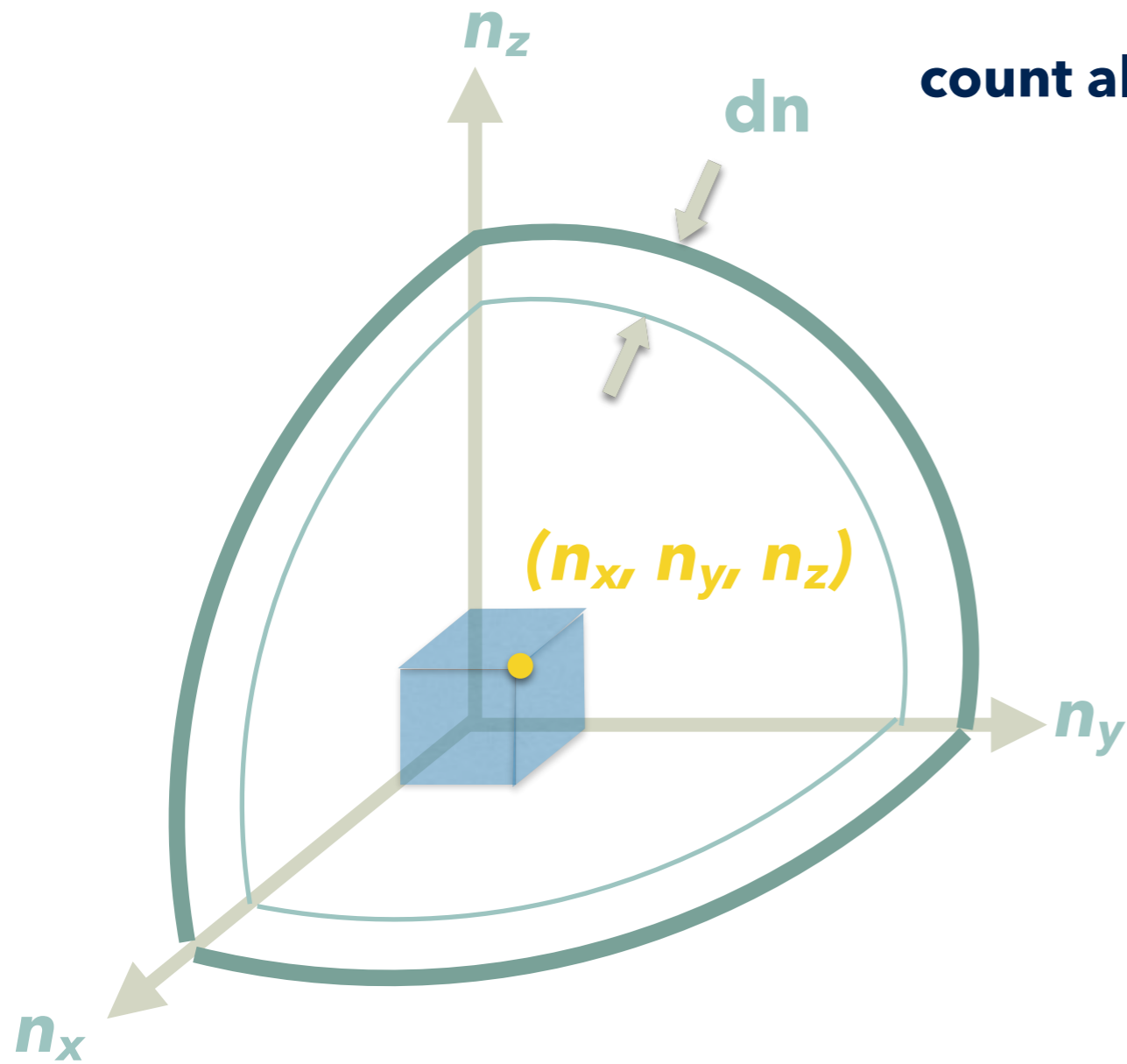
$$= \pi \left( \frac{L}{\pi c} \right)^3 \int_0^\infty \frac{\hbar\omega^3}{\exp \frac{\hbar\omega}{\tau} - 1} d\omega$$

$$u(\omega)d\omega = \frac{\pi}{\pi^3 c^3} \frac{\hbar\omega^3}{\exp \frac{\hbar\omega}{\tau} - 1} d\omega$$

$$= \frac{1}{\pi^2 c^3} \frac{\hbar\omega^3}{\exp \frac{\hbar\omega}{\tau} - 1} d\omega$$

we do not integrate

$$u = \frac{U}{V} = \frac{U}{L^3}$$



count all modes in phase space

$$\omega = \frac{n\pi c}{L}$$

$$\omega = 2\pi\nu$$

$$u(\nu) = u(\omega) \cdot 2\pi$$

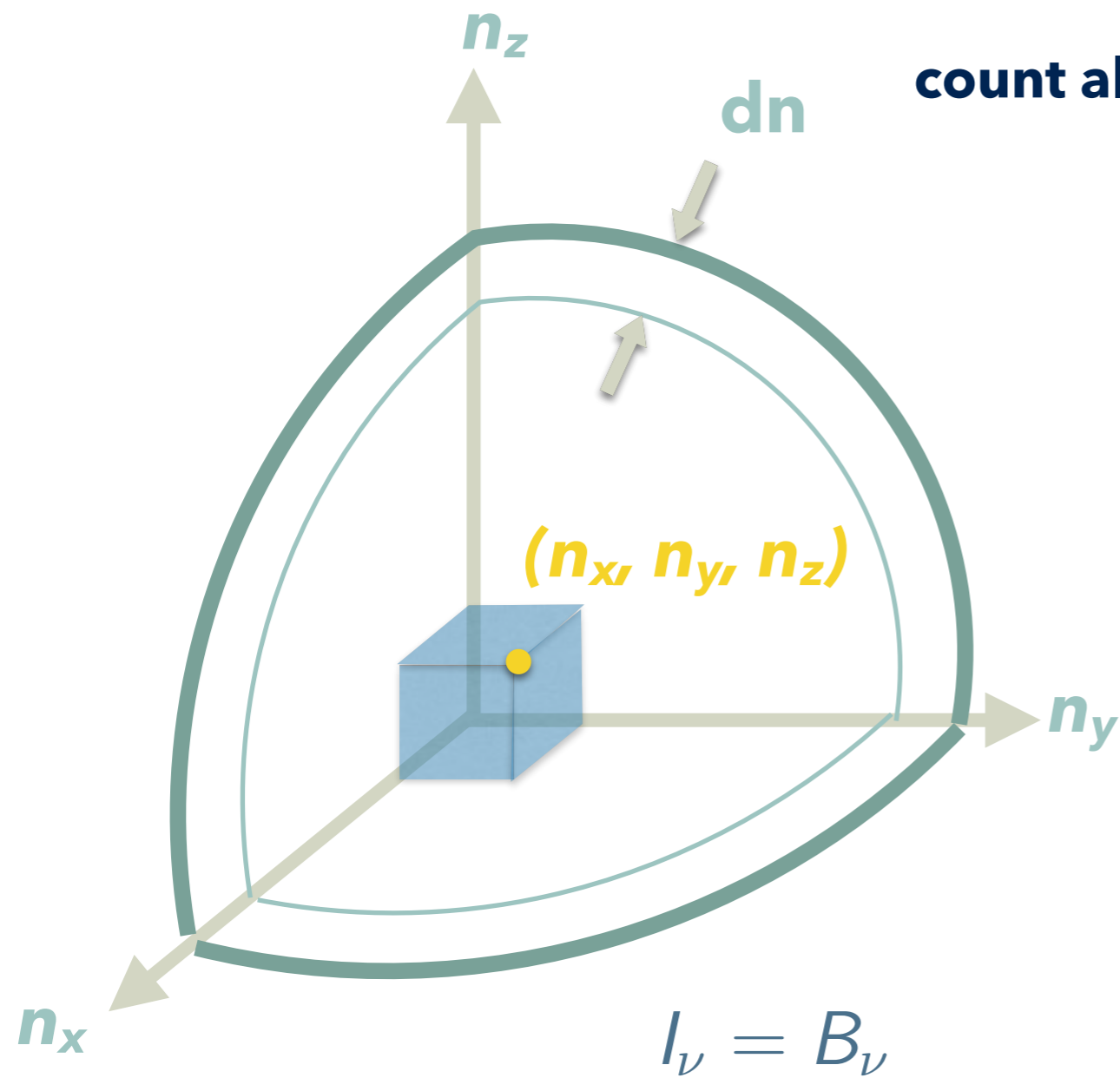
$$= \frac{1}{\pi^2 c^3} \frac{h(2\pi)^3 \nu^3}{\exp \frac{h\nu}{kT} - 1}$$

$$= \frac{8\pi}{c^3} \frac{h\nu^3}{\exp \frac{h\nu}{kT} - 1}$$

$$u(\omega) d\omega = \frac{\pi}{\pi^3 c^3} \frac{\hbar \omega^3}{\exp \frac{\hbar \omega}{\tau} - 1} d\omega$$

$$= \frac{1}{\pi^2 c^3} \frac{\hbar \omega^3}{\exp \frac{\hbar \omega}{\tau} - 1} d\omega$$





$$\omega = \frac{n\pi c}{L}$$

$$\omega = 2\pi\nu$$

$$u(\nu) = u(\omega) \cdot 2\pi$$

$$= \frac{1}{\pi^2 c^3} \frac{h(2\pi)^3 \nu^3}{\exp \frac{h\nu}{kT} - 1}$$

$$= \frac{8\pi}{c^3} \frac{h\nu^3}{\exp \frac{h\nu}{kT} - 1}$$

$$B(\nu) = \frac{c}{4\pi} u(\nu)$$

$$= \frac{c}{4\pi} \frac{8\pi}{c^3} \frac{h\nu^3}{\exp \frac{h\nu}{kT} - 1}$$

$$= \frac{2h\nu^3}{c^2} \frac{1}{\exp \frac{h\nu}{kT} - 1}$$

# Stefan-Boltzmann law

total energy from a unit surface?

$$B(\nu) = \frac{2h\nu^3}{c^2} \frac{1}{\exp\left(\frac{h\nu}{kT}\right) - 1}$$

Planck function

$$I_\nu = B_\nu$$

$$F_\nu = \pi I_\nu$$

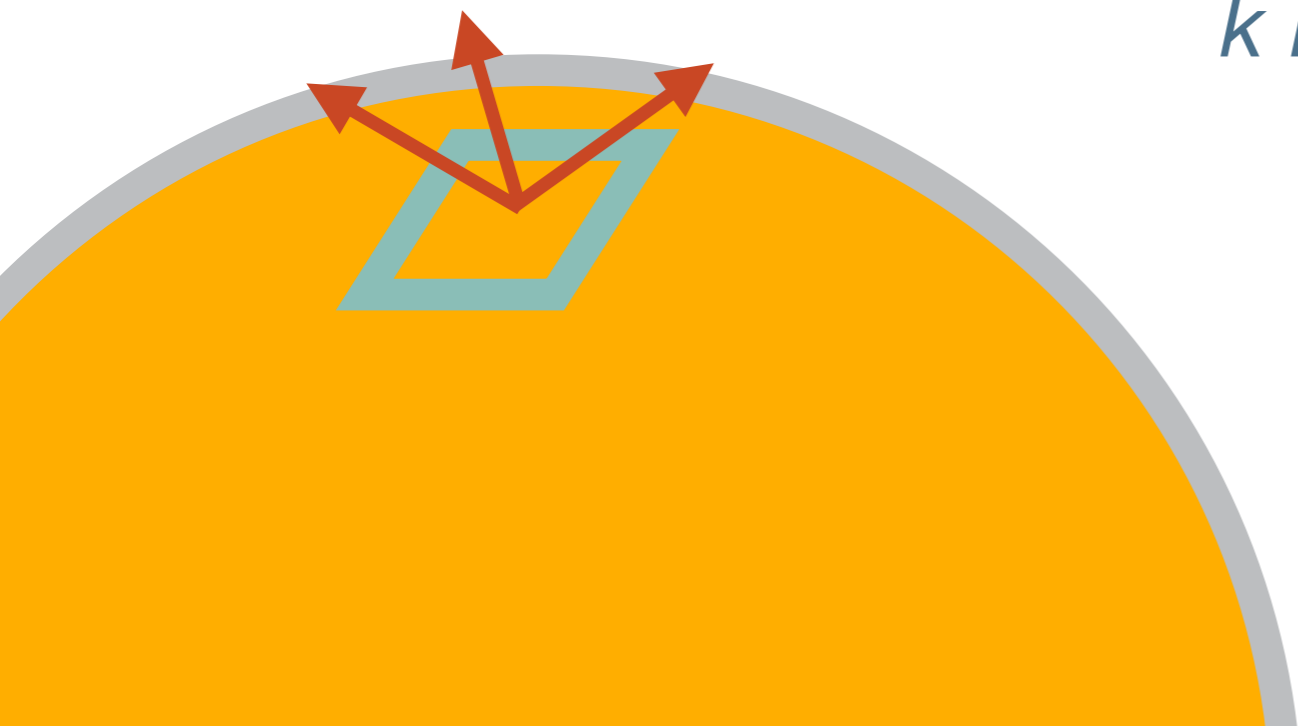
$$= \pi B_\nu$$

$$E = \int_0^\infty B_\nu d\nu$$

$$u = \frac{h\nu}{kT}$$

$$F = \sigma T^4$$

$$\int_0^\infty \frac{x^3}{e^x - 1} dx = \frac{\pi^4}{15}$$



**1** Stefan-Boltzmann constant

$$\sigma = \frac{2\pi^5 k^4}{15c^2 h^3}$$

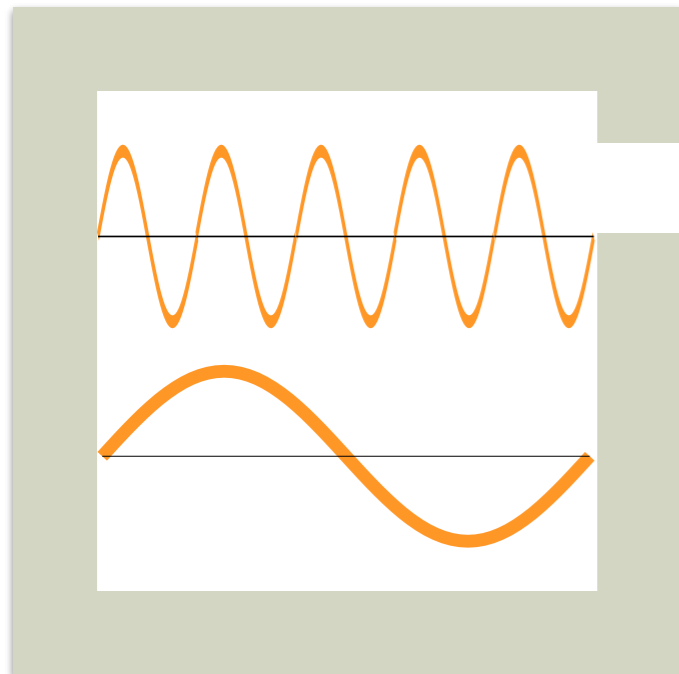
# Blackbody is

- 1** **optically thick** if not, call it a graybody

blackbody is not a shape of spectrum  
it is an absolute level

- 2** **strongest emission at given frequency**  
at given temperature

all possible modes are fully occupied  
cannot put photons any more



whatever the mechanism is,  
when you make an emission  
stronger and stronger,  
and reached  $B(T)$ ,  
that is the end.

# Planck function

you only see the peak of Blackbody

$$\lambda T = 3000 \text{ } [\mu\text{m} \cdot \text{K}]$$

10 K

300  $\mu\text{m}$  sub-mm

100 K

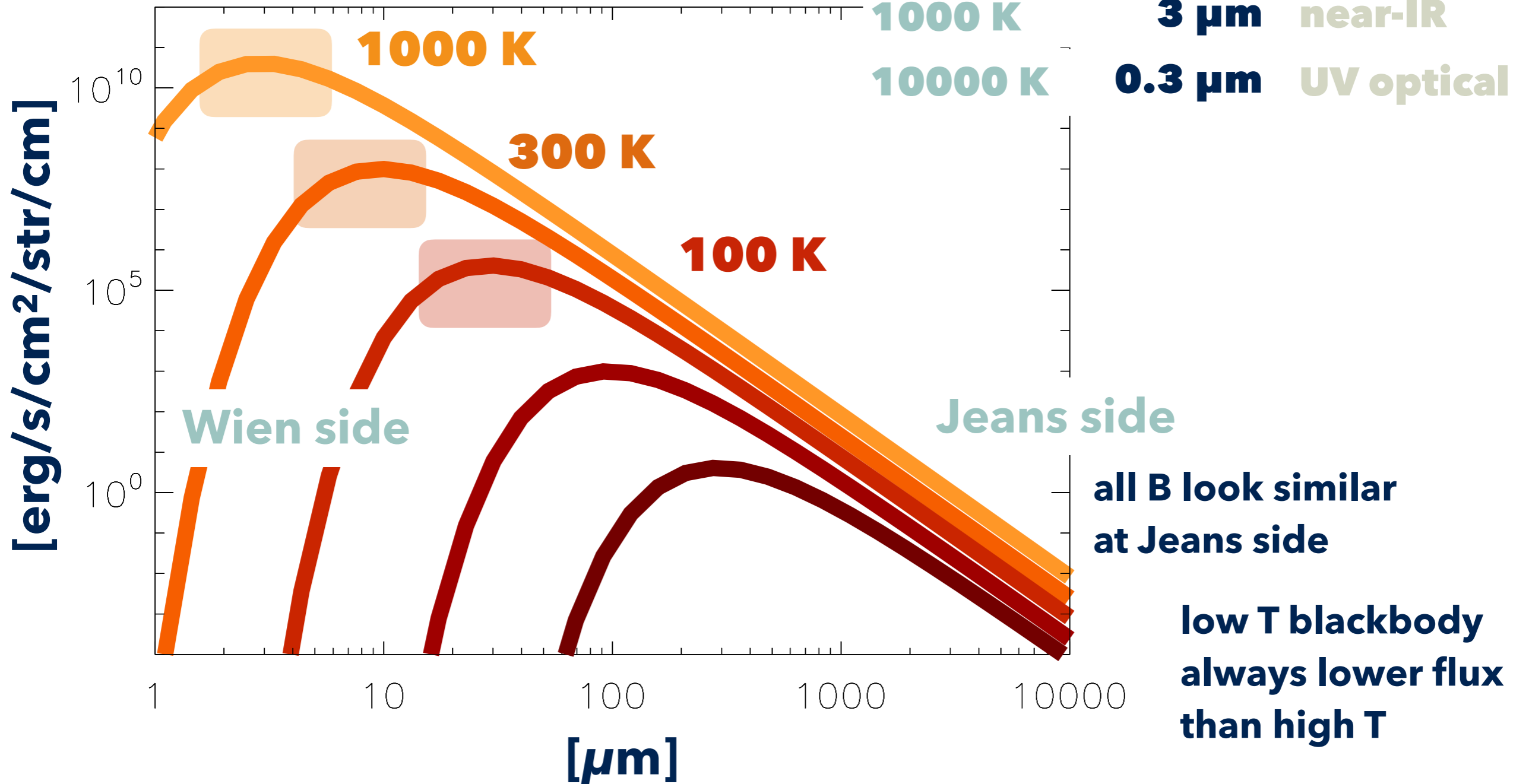
30  $\mu\text{m}$  mid-IR

1000 K

3  $\mu\text{m}$  near-IR

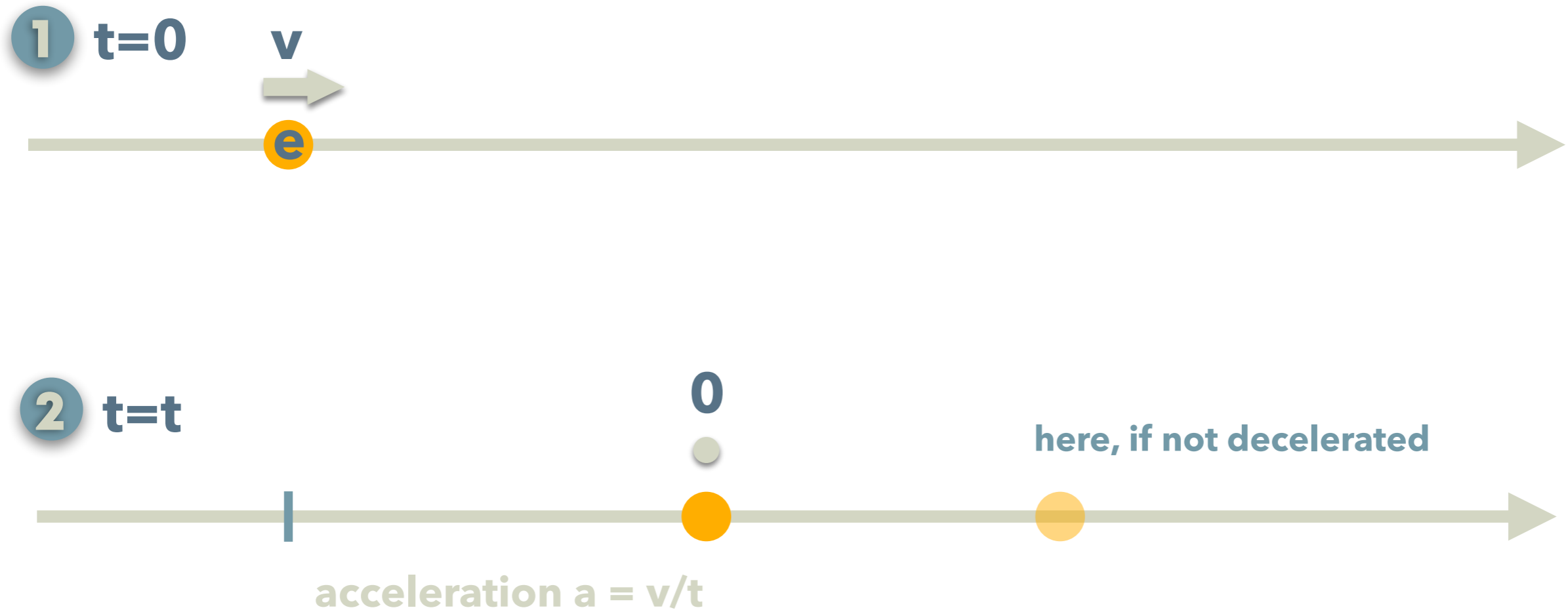
10000 K

0.3  $\mu\text{m}$  UV optical

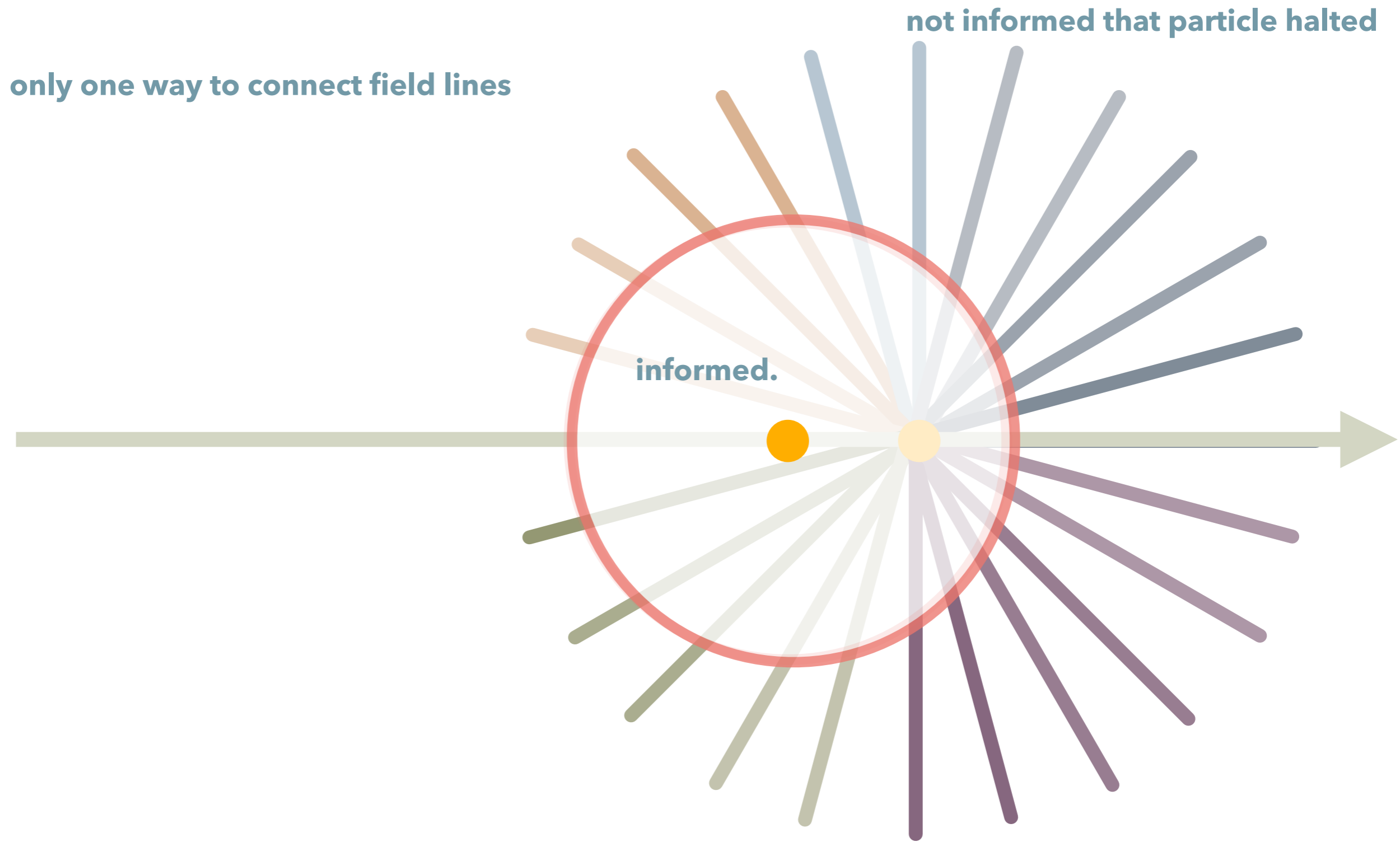


<b>10 K</b>	<b>300 <math>\mu\text{m}</math></b>	<b>sub-mm</b>	<b>gas, clouds dust</b>
<b>100 K</b>	<b>30 <math>\mu\text{m}</math></b>	<b>mid-IR</b>	<b>molecular torus of AGN protoplanetary disk</b>
<b>1000</b>	<b>3 <math>\mu\text{m}</math></b>	<b>near-IR</b>	<b>planet brown dwarfs</b>
<b>10000 K</b>	<b>0.3 <math>\mu\text{m}</math></b>	<b>UV optical</b>	<b>reflected light of stars asteroids, planets stars (transit planet) stars (radial velocity)</b>

# Radiation



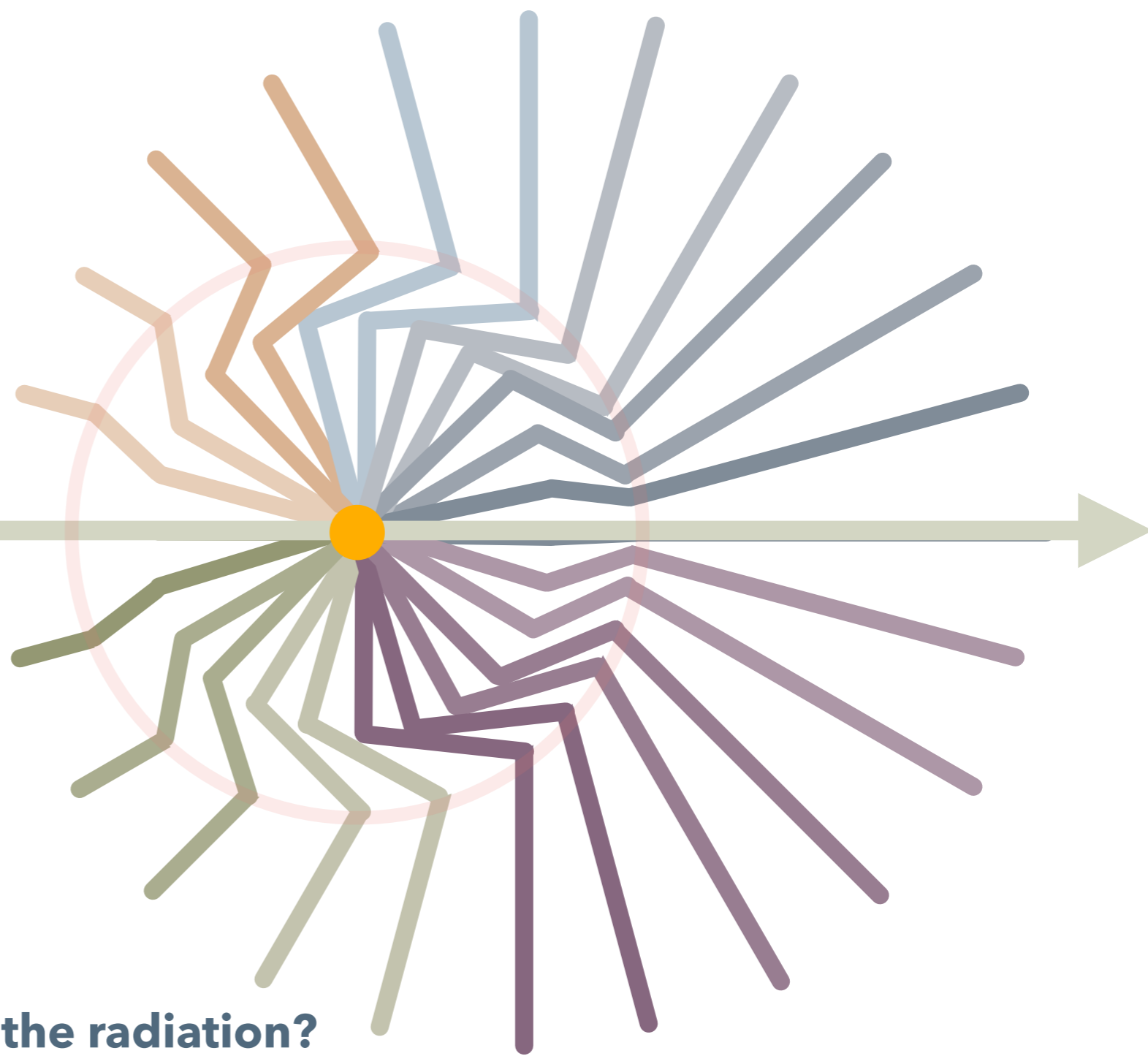






only one way to connect field lines

- 1 how much is the thickness of the transition zone ?
- 2 direction of the electric field?
- 3 how this transition zone develops with time?
- 4 explain intensity of electric field decays as  $1/r$  instead of  $1/r^2$
- 5 where is the strongest part of the radiation?



## What we can take from the cartoon

# Radiation

- 1**  $E$  is proportional to  $\dot{u}$
- 2**  $E$  is proportional to  $q$
- 3** radiates perpendicular to  $\dot{u}$

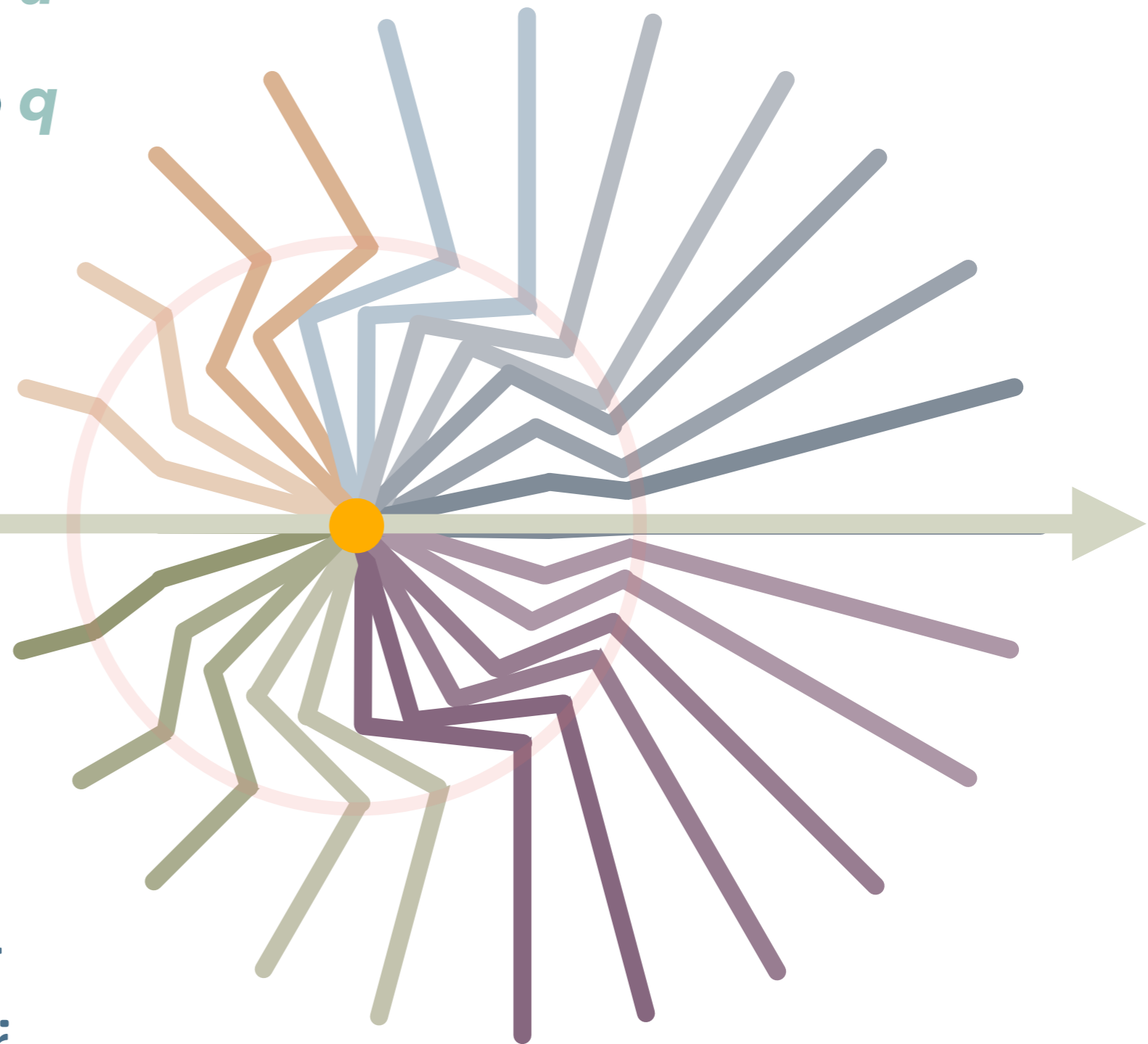
## Dipole approximation

$$\frac{dP}{d\Omega} = \frac{\ddot{\mathbf{d}}^2}{4\pi c^3} \sin^2 \Theta$$

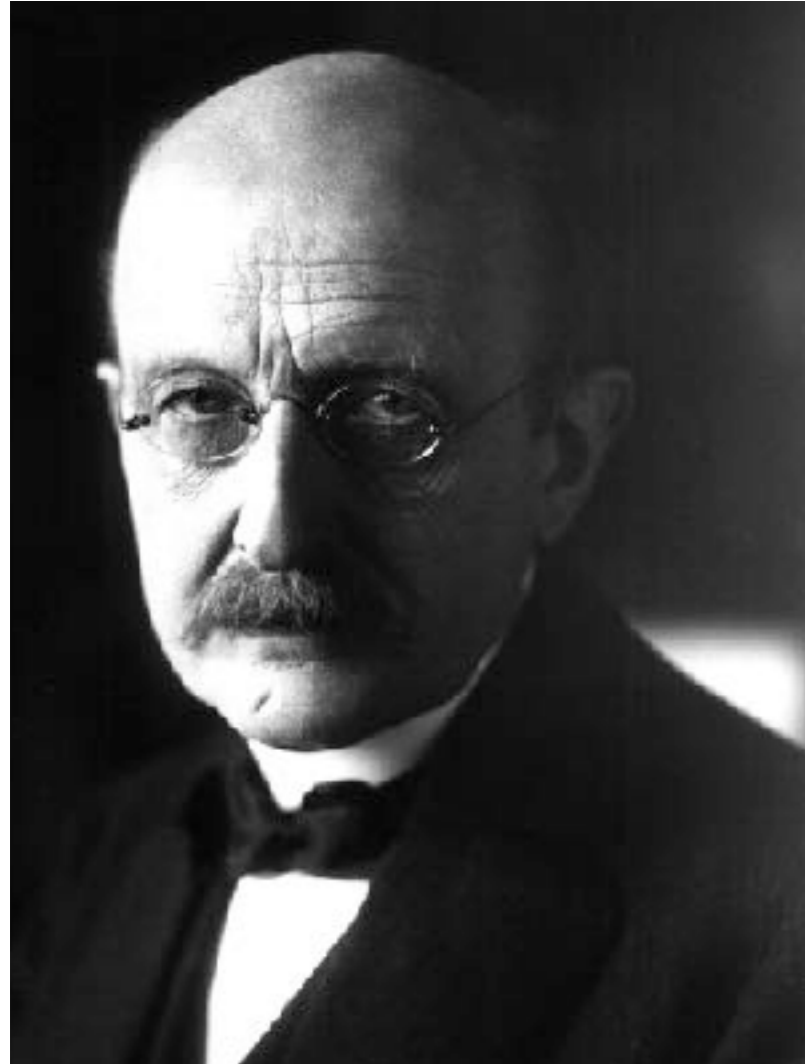
Dipole moment

$$\mathbf{d} = e\mathbf{r}$$

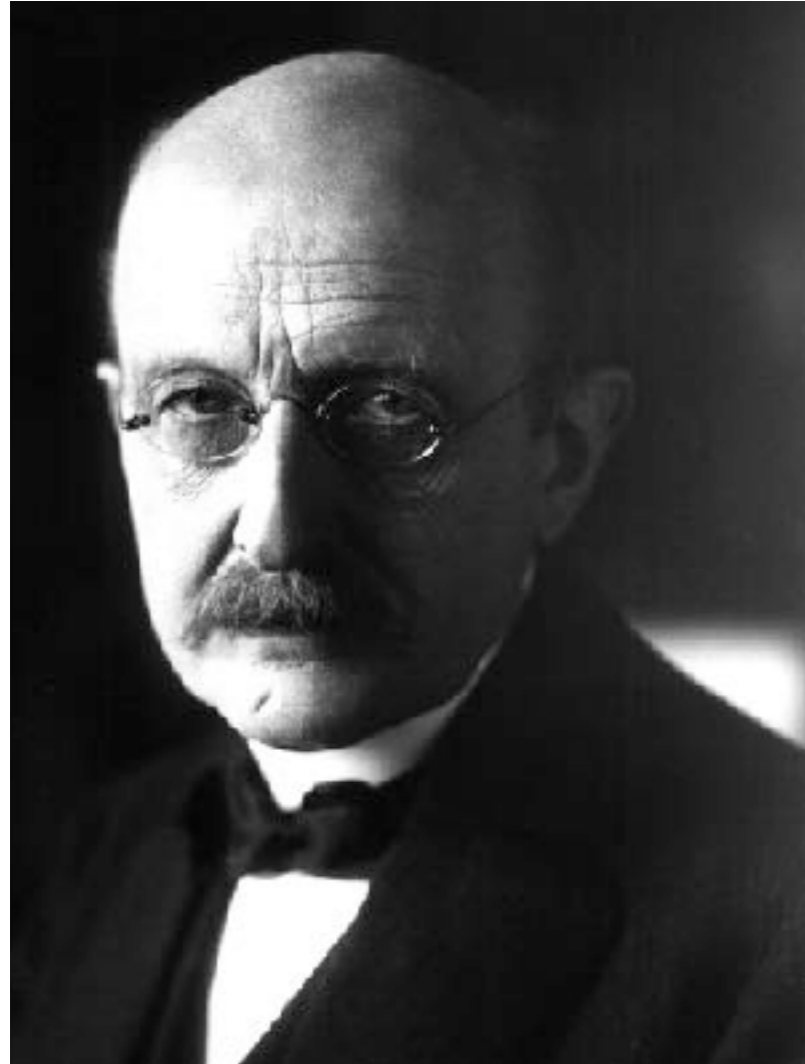
$$\ddot{\mathbf{d}} = e\ddot{\mathbf{r}}$$



# Thomson scattering



# Thomson scattering



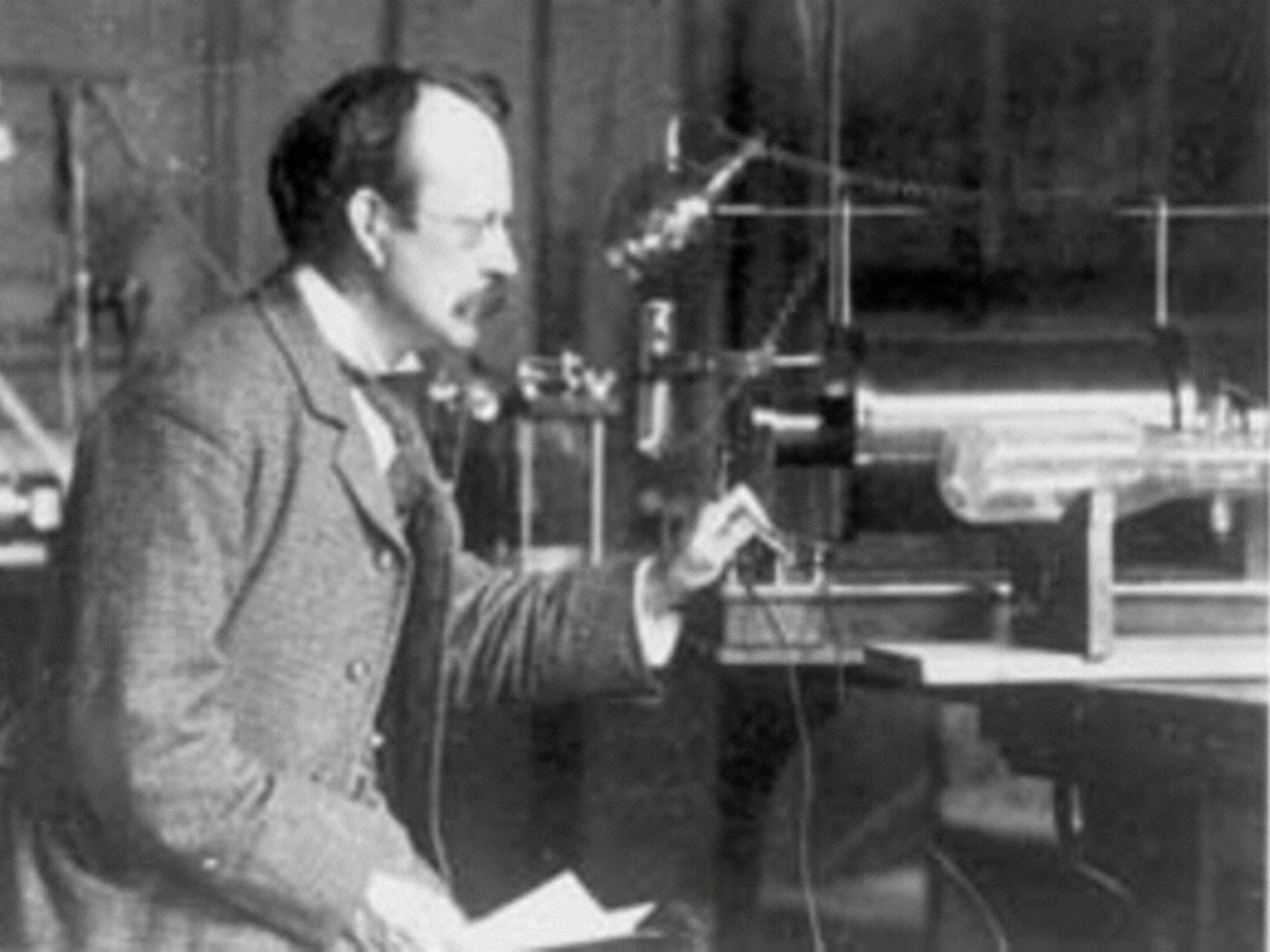
**Max Planck**

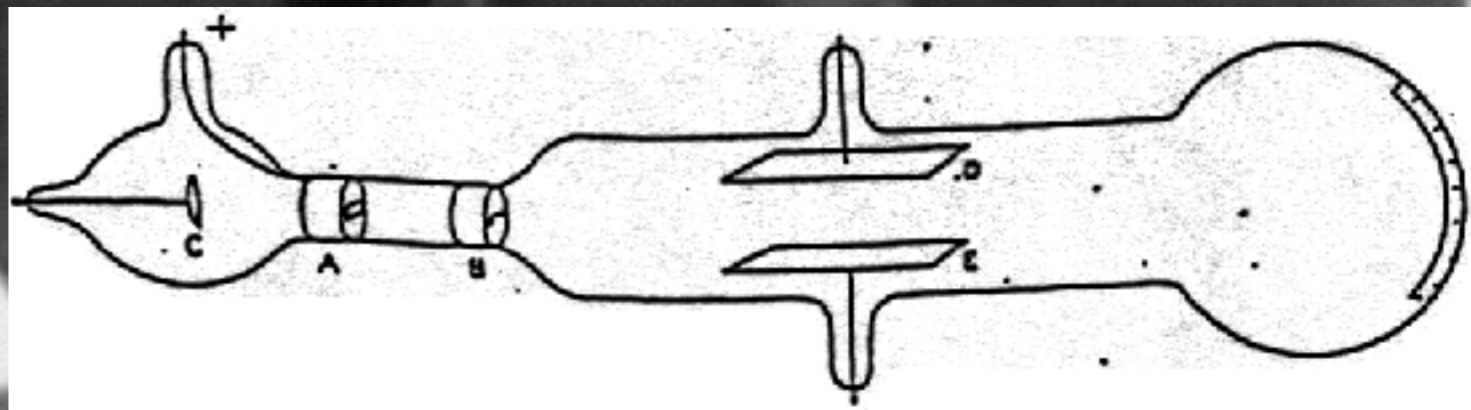
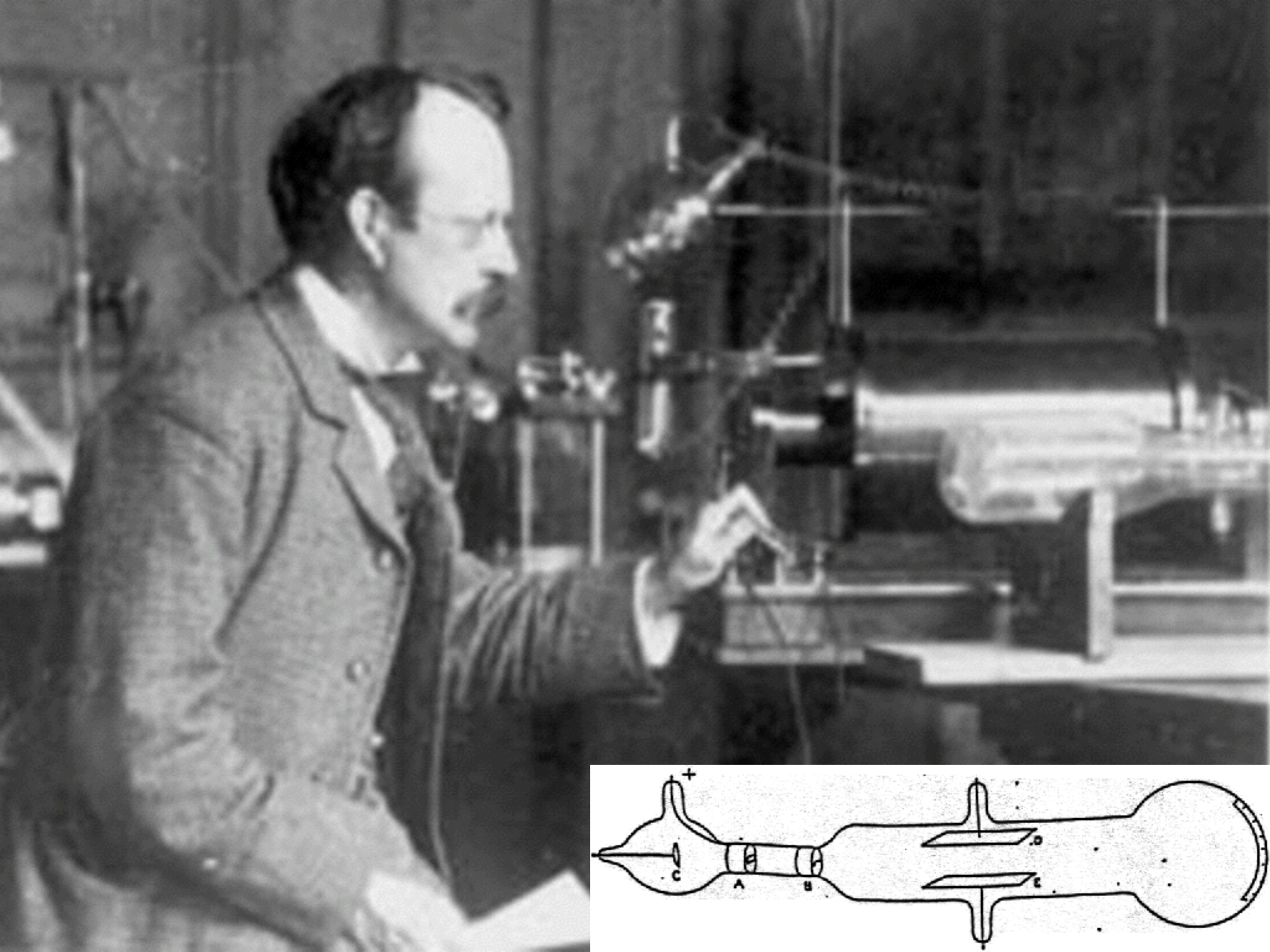


**J. J. Thomson**



**David Hilbert**





# Where is the acceleration comes from?

something that pushes and pulls a charge?

**Electromagnetic field!**

low-energy photon

**Thomson scattering**



**free-charge**

**Rayleigh scattering**



**bound charge**

**Compton scattering**



**free-charge**

high-energy photon

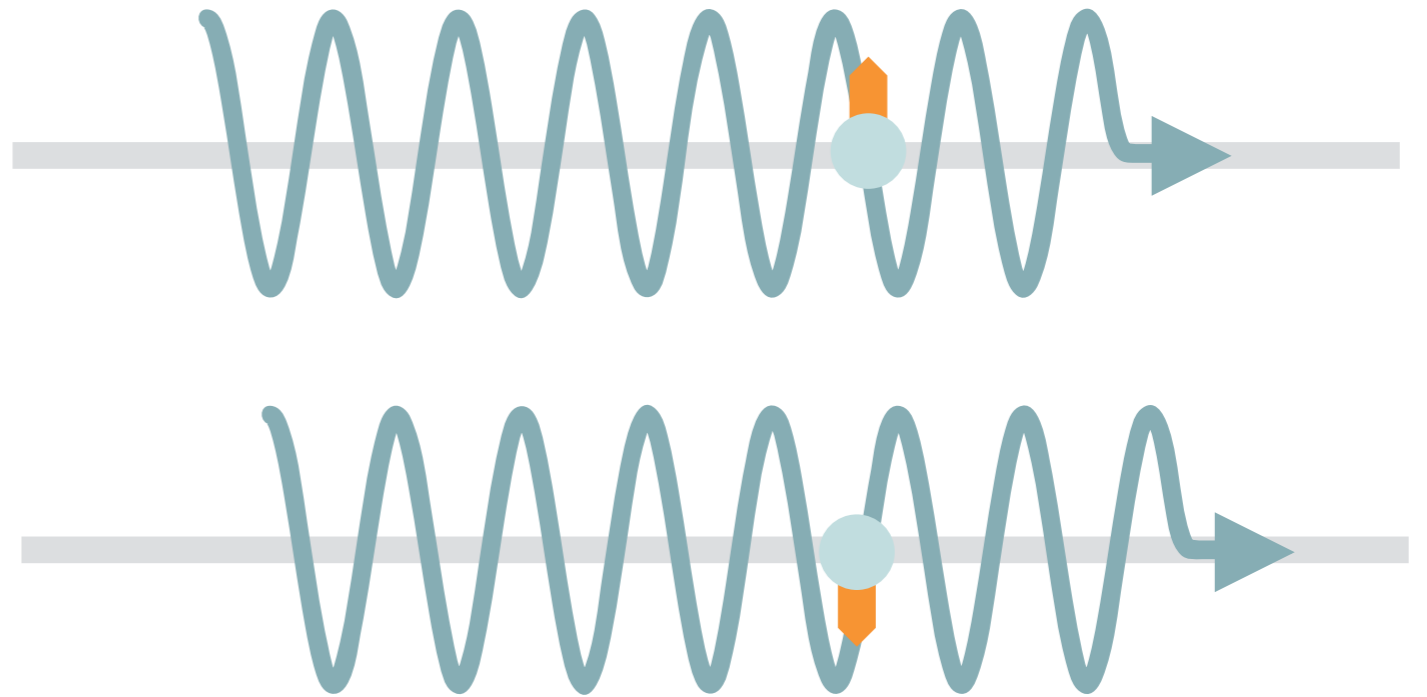


**Rayleigh scattering**



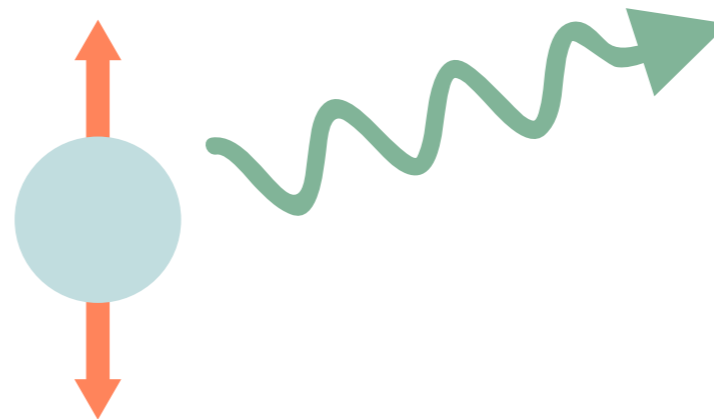
## Dipole approximation

- 1**  $E$  is proportional to  $\dot{u}$
- 2**  $E$  is proportional to  $q$
- 3** radiates perpendicular to  $\dot{u}$



## Power

$$\frac{dP}{d\Omega} = \frac{\ddot{\mathbf{d}}^2}{4\pi c^3} \sin^2 \Theta$$



## Dipole approximation

$$\langle \sin^2 \omega t \rangle = \frac{1}{2}$$

$$\frac{dP}{d\Omega} = \frac{\ddot{\mathbf{d}}^2}{4\pi c^3} \sin^2 \Theta$$

## incoming electric field

$$\mathbf{E} = \boldsymbol{\varepsilon} E_0 \sin \omega t$$

$$m\ddot{\mathbf{r}} = e\mathbf{E}$$



## dipole moment

$$\mathbf{d} = e\mathbf{r}$$

$$\ddot{\mathbf{d}} = e\ddot{\mathbf{r}}$$

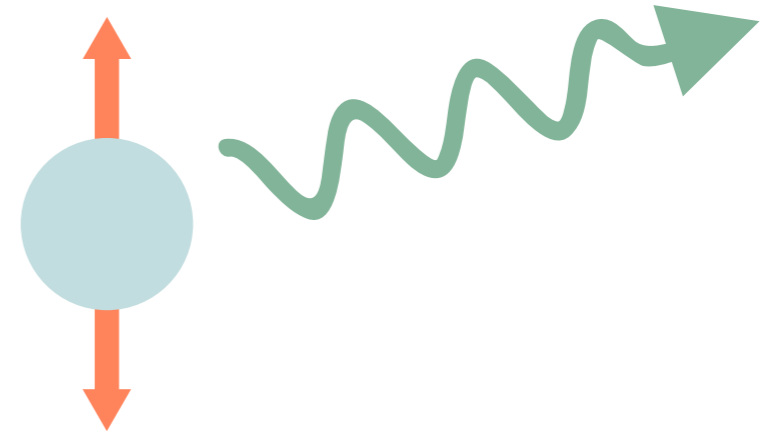
$$\ddot{\mathbf{d}} = \frac{e^2 \mathbf{E}}{m}$$

$$|\ddot{\mathbf{d}}|^2 = \frac{e^4 E_0^2}{2m^2}$$

## Dipole approximation

$$|\ddot{\mathbf{d}}|^2 = \frac{e^4 E_0^2}{2m^2}$$

$$\frac{dP}{d\Omega} = \frac{\ddot{\mathbf{d}}^2}{4\pi c^3} \sin^2 \Theta$$



## total energy scattered?

$$P = \int \frac{dP}{d\Omega} d\Omega$$

$$\frac{dP}{d\Omega} = \frac{e^4 E_0^2}{8\pi m^2 c^3} \sin^2 \Theta$$

$$P = \frac{e^4 E_0^2}{8\pi m^2 c^3} \cdot 2\pi \frac{4}{3}$$

$$= \frac{e^4 E_0^2}{3m^2 c^3}$$

$$\begin{aligned} \int \sin^2 \Theta d\Omega &= \int_0^{2\pi} \int_0^\pi \sin^2 \Theta \sin \Theta d\Theta d\Phi \\ &= 2\pi \int_0^\pi (1 - \cos^2 \Theta) \sin \Theta d\Theta \end{aligned}$$

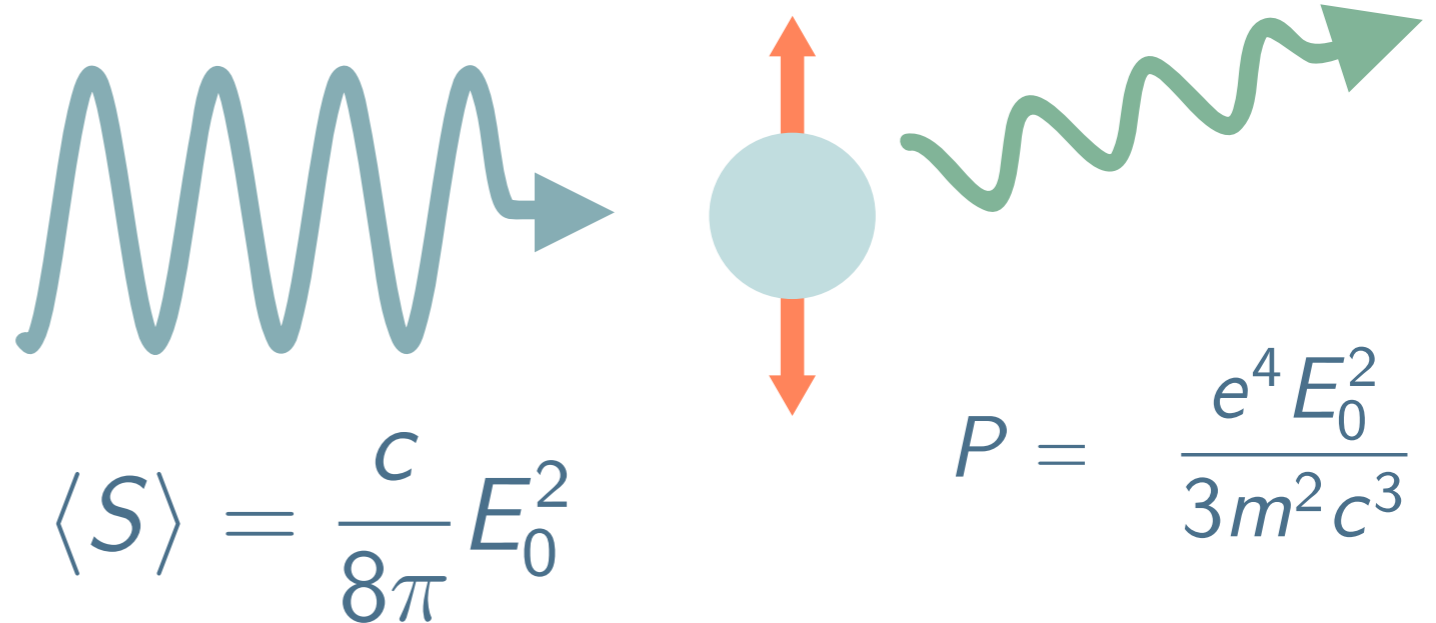
$$\begin{aligned} \int_0^\pi (1 - \cos^2 \Theta) \sin \Theta d\Theta &= -[\cos \Theta]_0^\pi + \left[ \frac{\cos^3 \Theta}{3} \right]_0^\pi \\ &= \left[ -(-1 - 1) + \frac{-1}{3} - \frac{1}{3} \right] \\ &= \frac{4}{3} \end{aligned}$$

## Thomson cross section

$$P = \langle S \rangle \sigma_T$$

$$\sigma_T = \frac{P}{\langle S \rangle}$$

$$= \frac{8\pi}{3} \frac{e^4}{m^2 c^4}$$



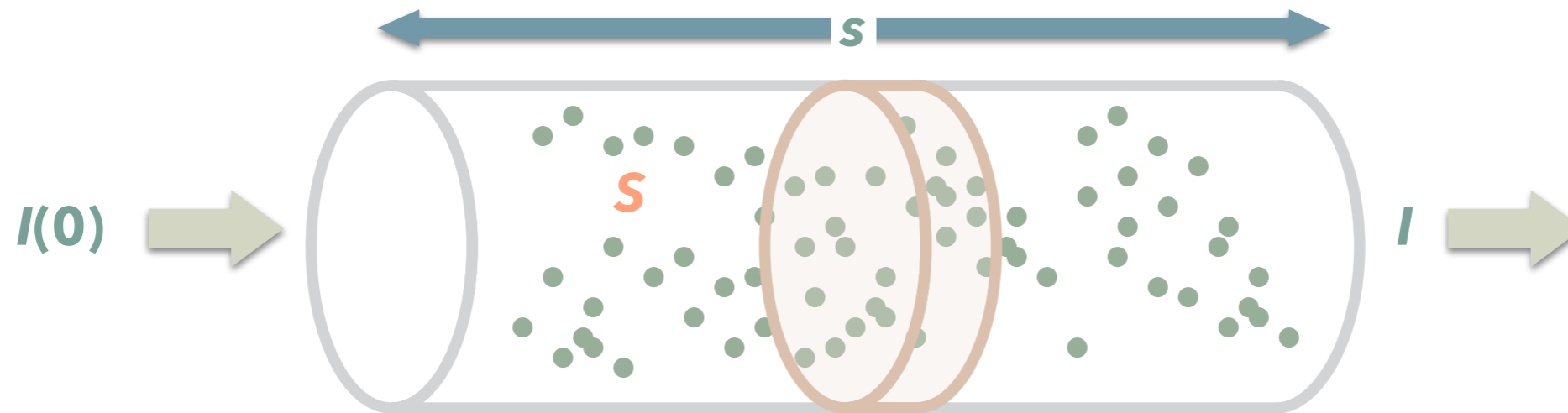
Poynting vector

~specific intensity  $I_\nu$

## Thomson scattering

- 1 Scattering is a re-emission
  - 2 Frequency independent
  - 3 Polarized as in the incident light
  - 4 Forward scattering
- 2 Why Thomson scattering is sometimes referred to "electron" scattering?

# Radiation transfer



source function

$ds$

we want to find exit  $I$

$$I = \int j ds \quad \text{why not?}$$

$$= S(1 - e^{-\tau}) \quad (\text{when no bg})$$

$$\frac{dI}{ds} = -aI + j$$

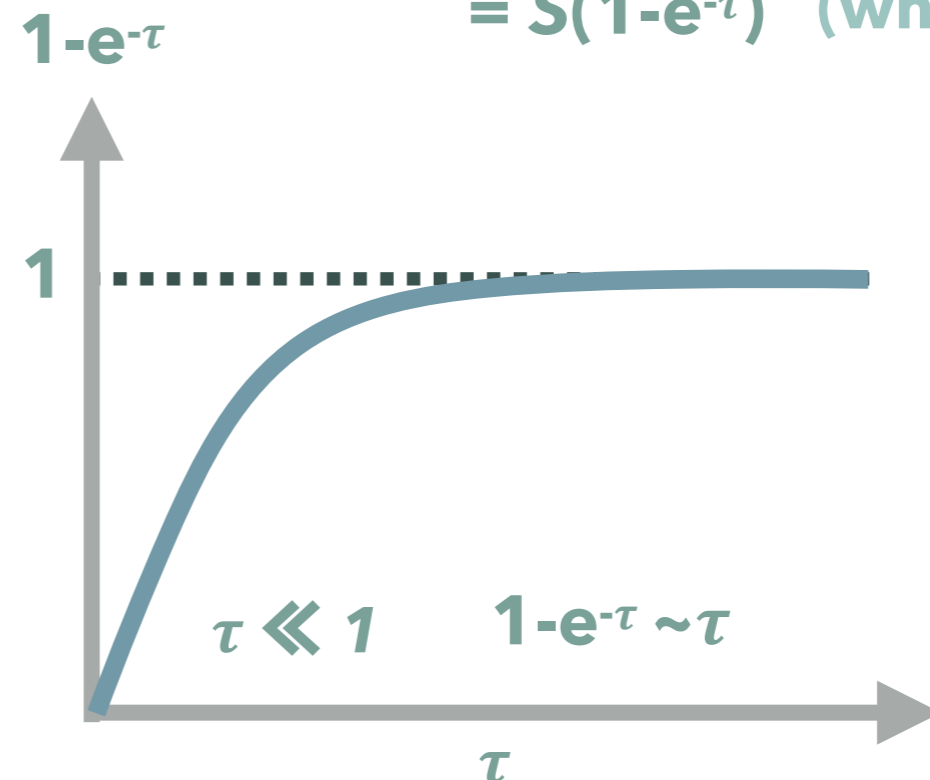
absorption coefficient  
emission coefficient

$$ads = d\tau \quad \text{optical depth}$$

$$S = \frac{j}{a} \quad \text{source function}$$

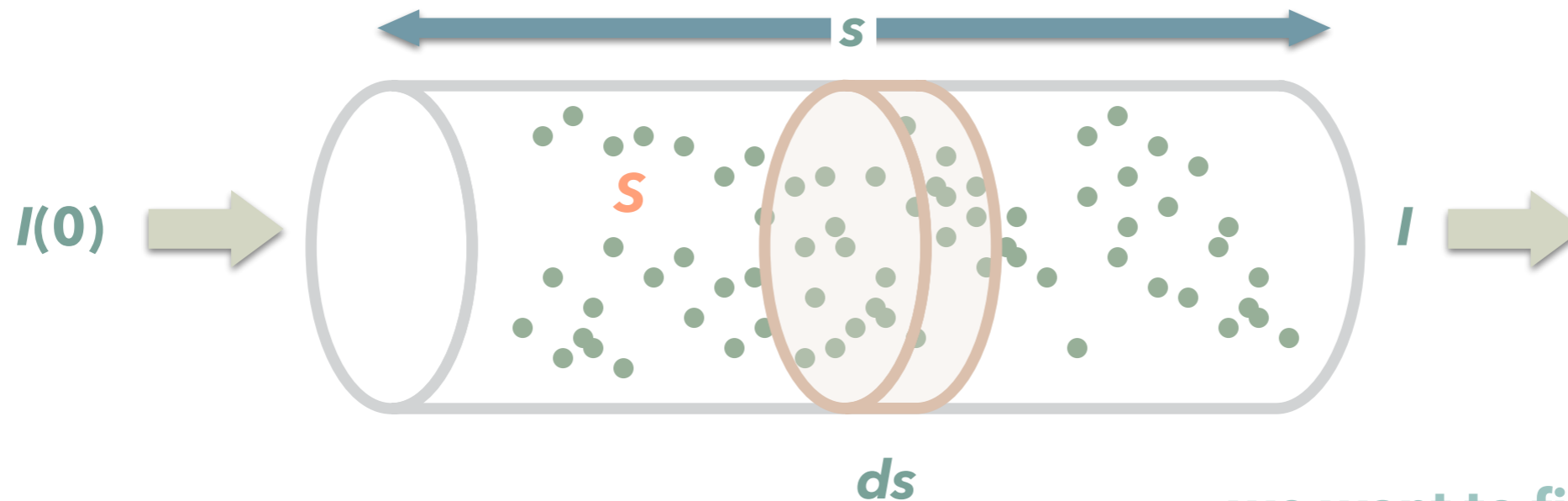
$$\frac{dI}{d\tau} = -I + S$$

$$I = I(0) e^{-\tau} + S(1 - e^{-\tau})$$



**3** Find above.

# Radiation transfer



$as = \tau$       **optical depth**

$S = \frac{j}{a}$       **source function**

$S(1 - e^{-\tau}) = S\tau$

$= \frac{j}{a} \tau$

$= \frac{j}{a} as$

$= j s$

optically thin case  
indeed  $I = j \cdot ds$

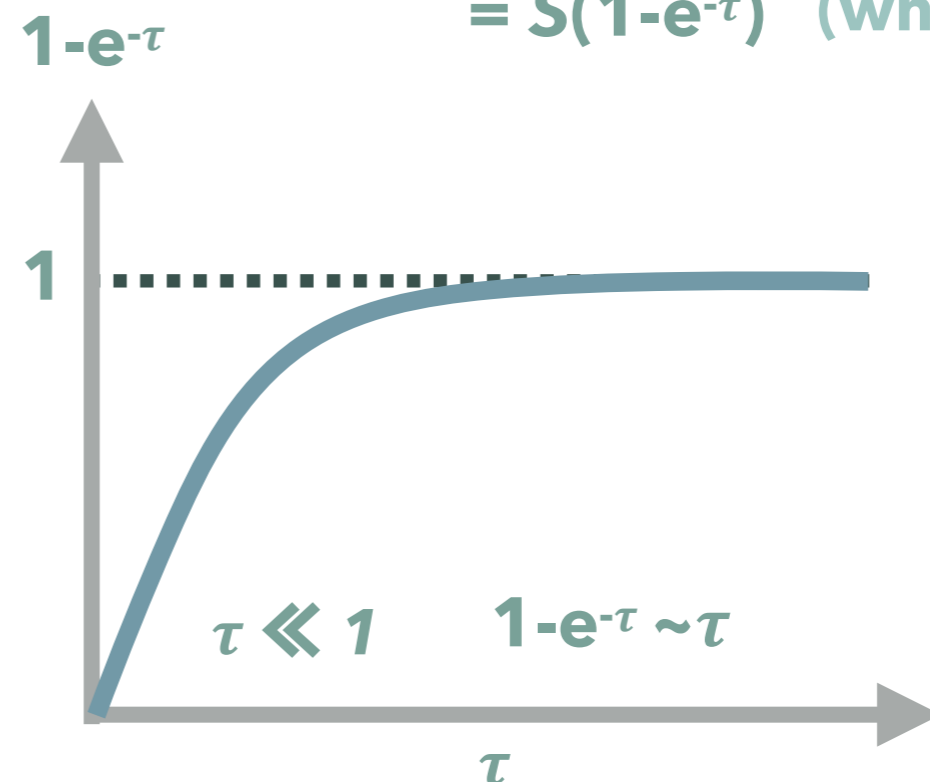
$a = n\sigma$

**source function** :  $j$  normalized by matter

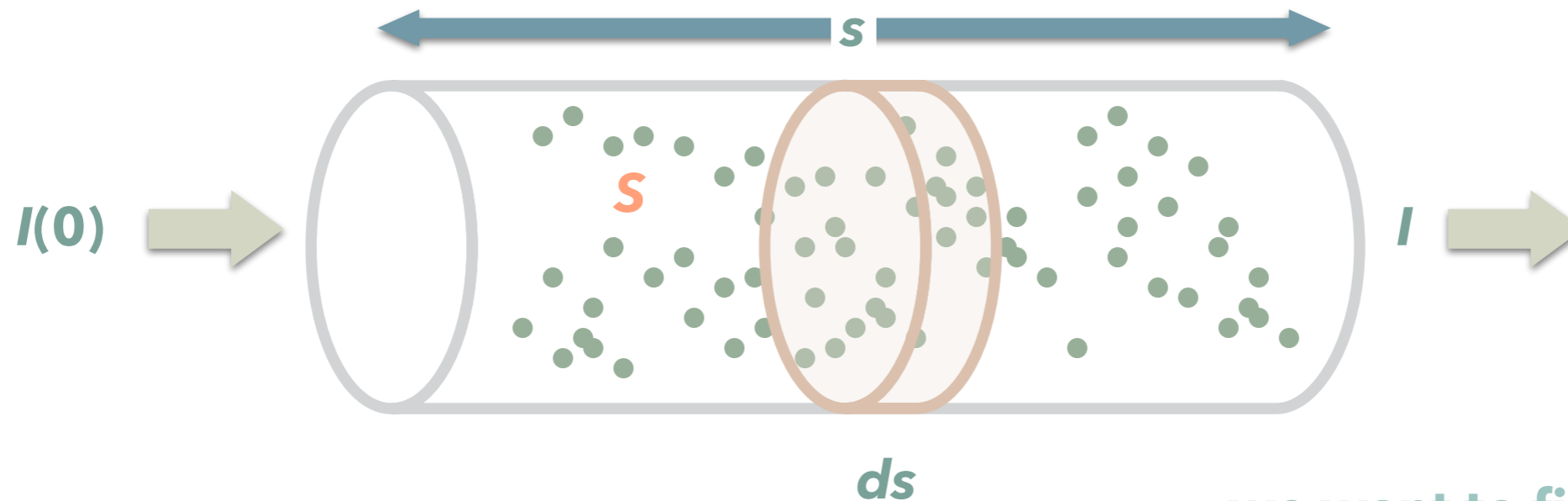
we want to find exit  $I$

$I = \int j ds$       why not?

$= S(1 - e^{-\tau})$  (when no bg)



# Radiation transfer



$as = \tau$       **optical depth**

$S = \frac{j}{a}$       **source function**

$= B$       **blackbody**  
**Kirchhoff's law**

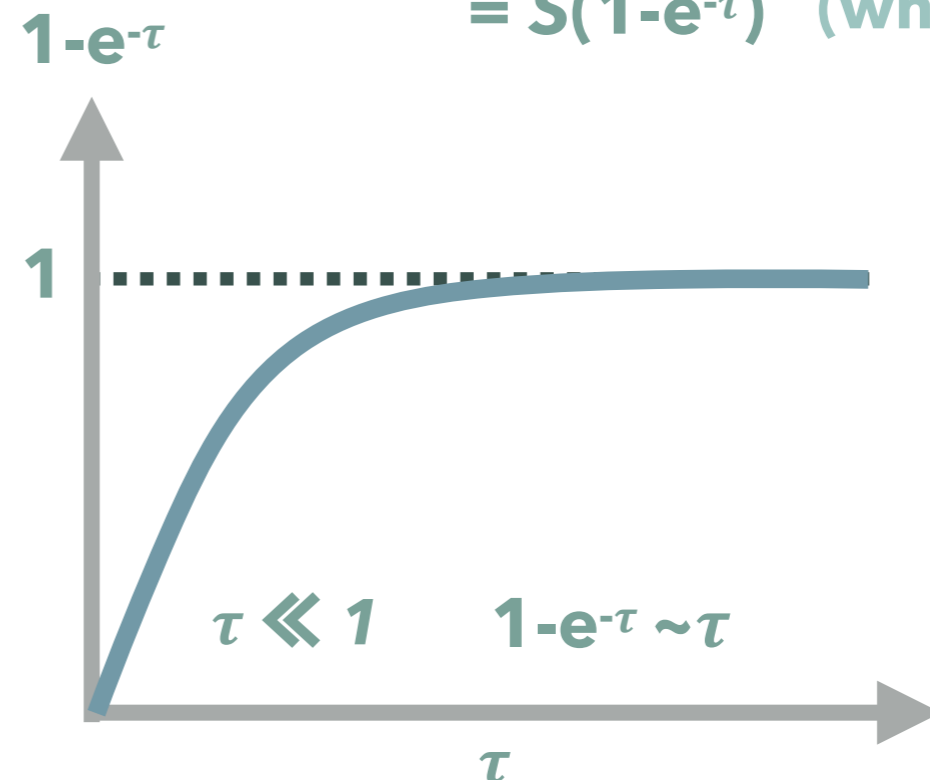
$$I = I(0) e^{-\tau} + S(1 - e^{-\tau})$$

exit intensity  
 never overcomes blackbody

we want to find exit  $I$

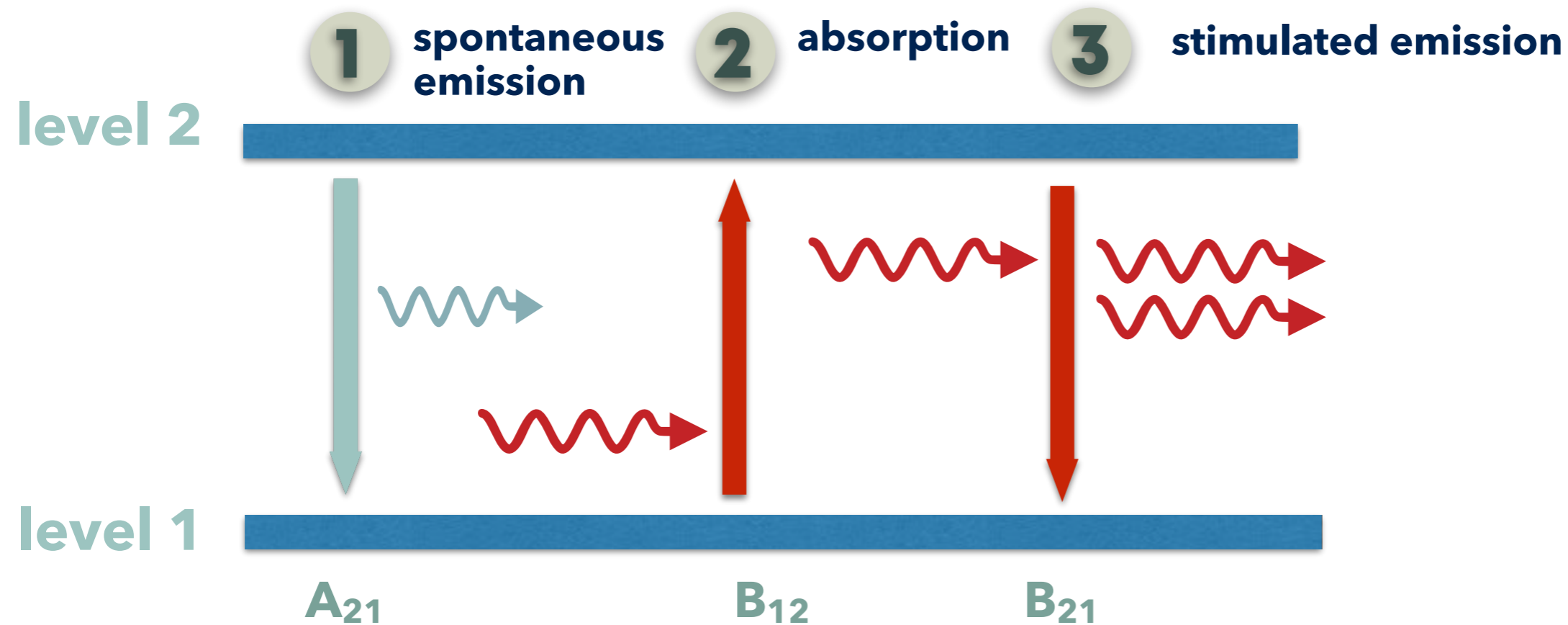
$$I = \int j ds \quad \text{why not?}$$

$$= S(1 - e^{-\tau}) \quad (\text{when no bg})$$

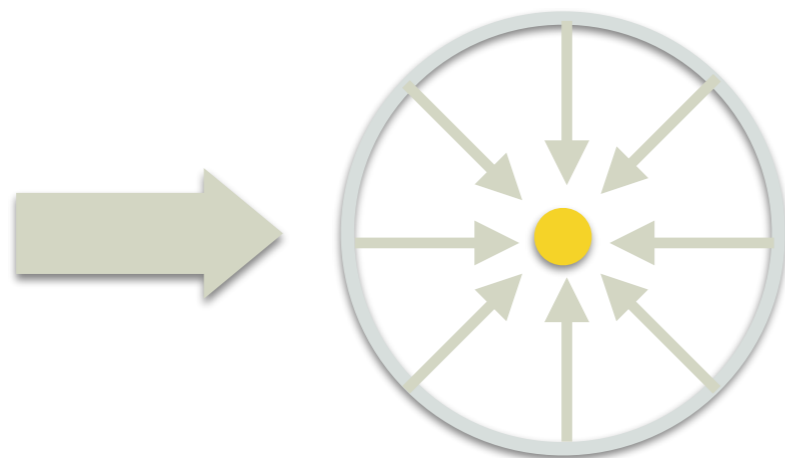


# Einstein coefficient

how a molecule interact with radiation?



$$J = \frac{1}{4\pi} \int I \, d\Omega$$



averaged out

system we are talking about

no collision

radiation only

two levels

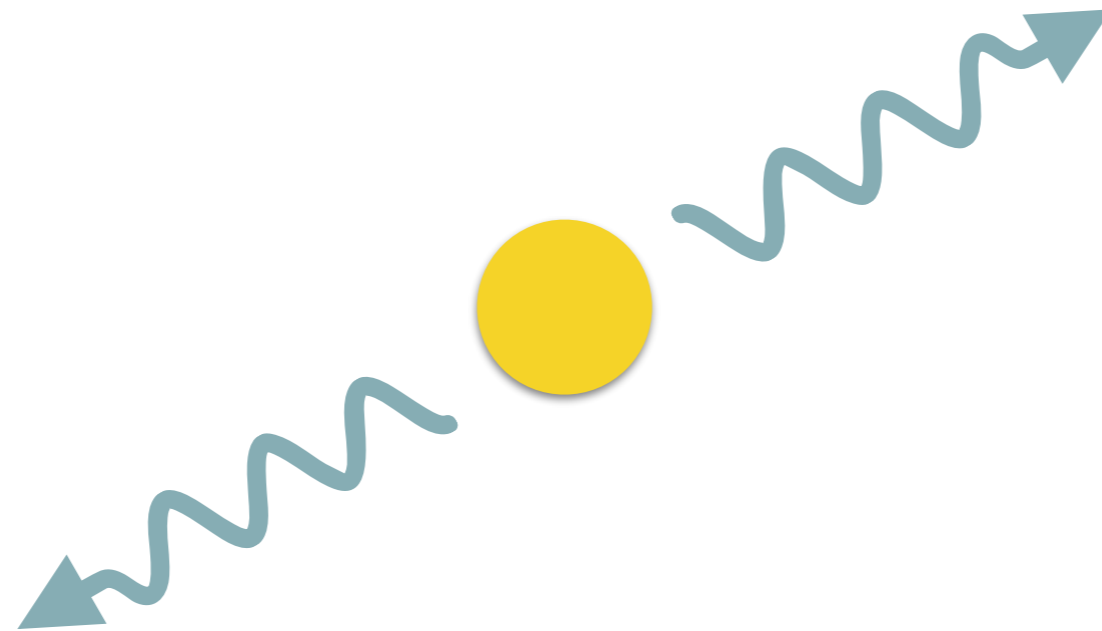
thermal equilibrium



a molecule or an atom in level 2

$A_{21}$  spontaneous  
emission  
[s<sup>-1</sup>]

you wait for one second.

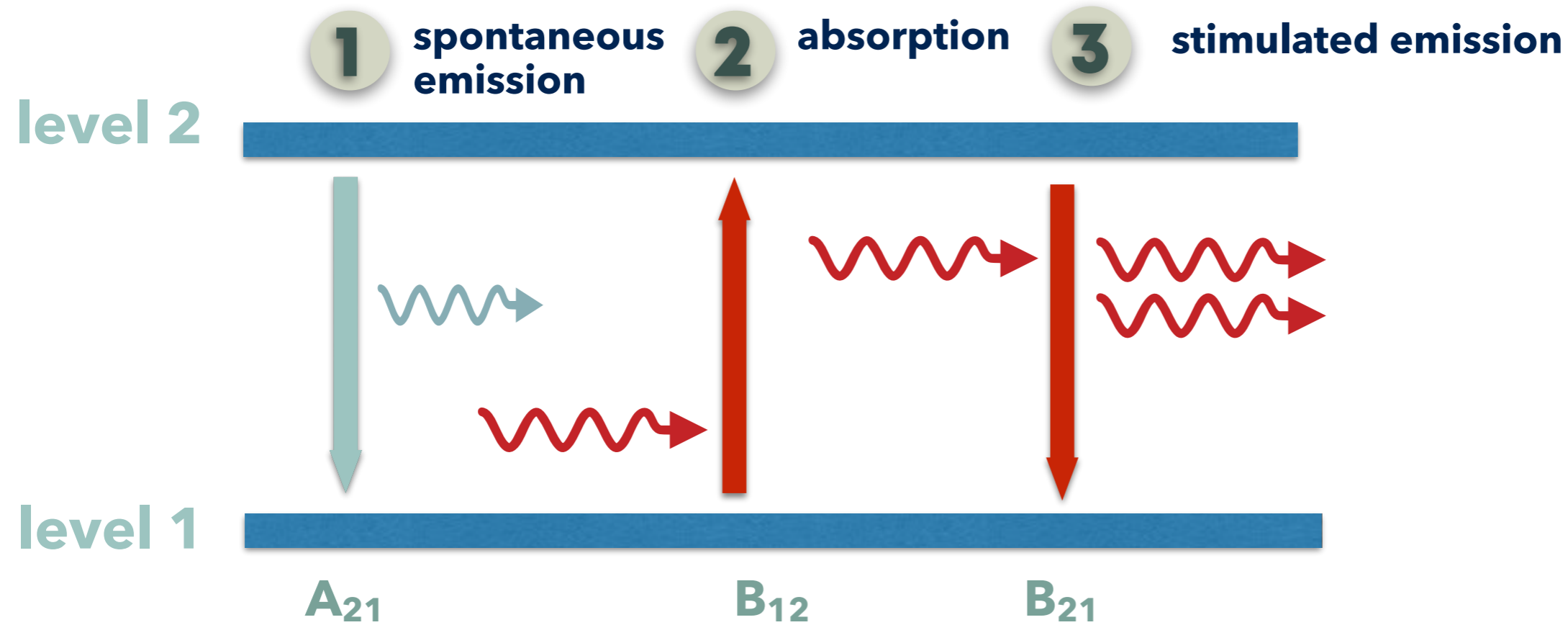


how often atoms /  
molecule decays to  
level 1

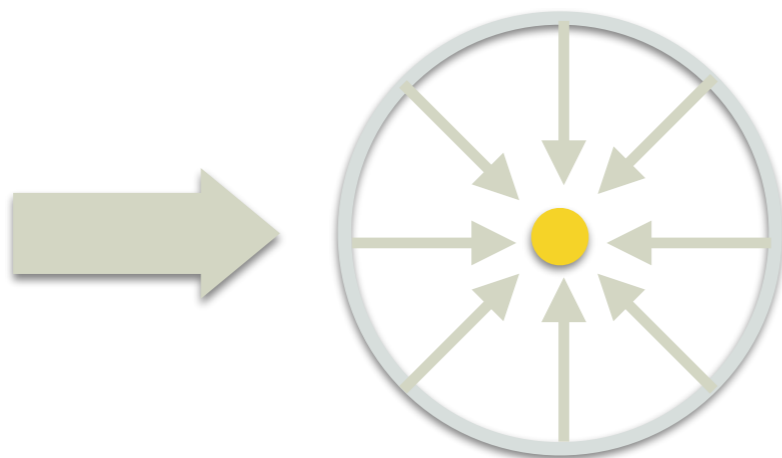
note it decays to  $4\pi$

# Einstein coefficient

how a molecule interact with radiation?



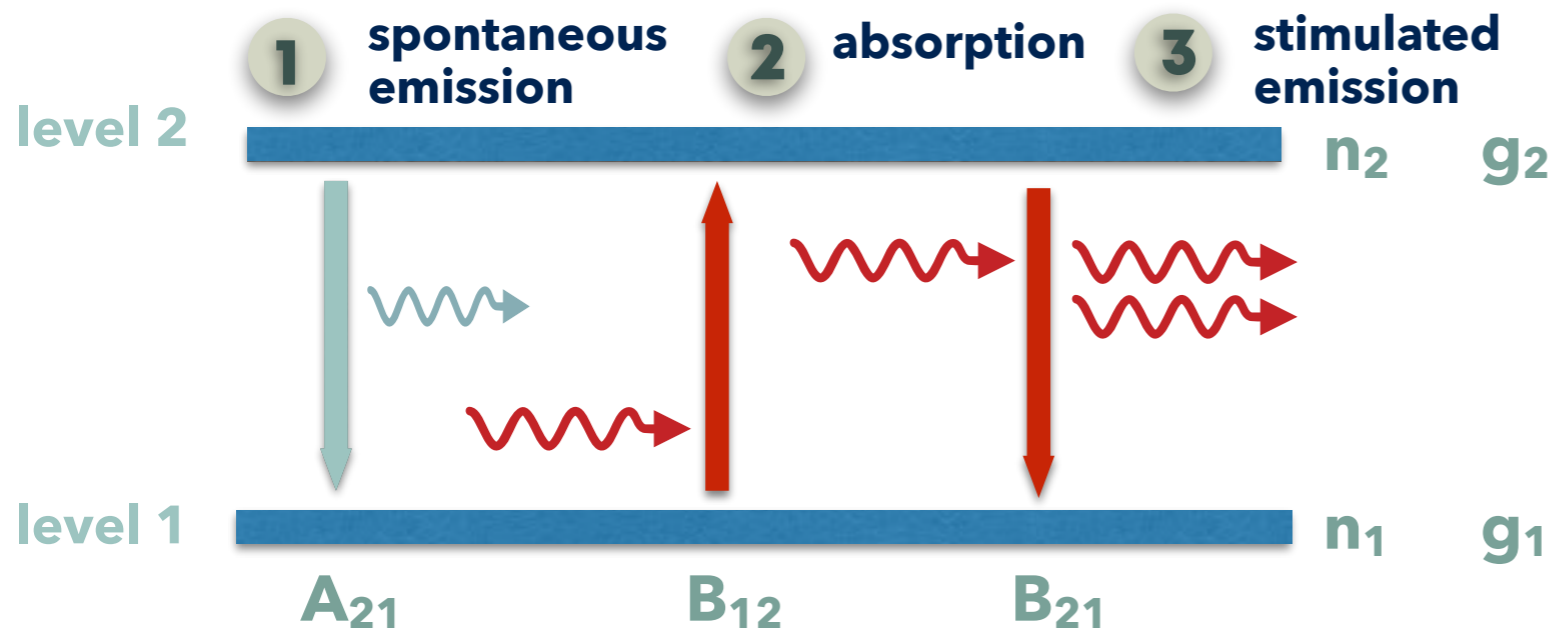
$$J = \frac{1}{4\pi} \int I \, d\Omega$$



averaged out

up                      down

$Jn_1B_{12} = n_2A_{21} + Jn_2B_{21}$



$$J n_1 B_{12} = J n_2 B_{21} + n_2 A_{21}$$

$$J (n_1 B_{12} - n_2 B_{21}) = n_2 A_{21}$$

$$J = \frac{n_2 A_{21}}{n_1 B_{12} - n_2 B_{21}}$$

$$J = \frac{\frac{n_2 A_{21}}{n_2 B_{21}}}{\frac{n_1 B_{12}}{n_2 B_{21}} - 1}$$

$$J = \frac{\frac{n_2 A_{21}}{n_2 B_{21}}}{\frac{g_1 B_{12}}{g_2 B_{21}} \exp\left(\frac{\epsilon}{\tau}\right) - 1}$$

$$B = \frac{2h\nu^3}{c^2} \frac{1}{\exp\left(\frac{h\nu}{kT}\right) - 1}$$

$$\frac{g_1 B_{12}}{g_2 B_{21}} = 1$$

$$\frac{A_{21}}{B_{21}} = \frac{2h\nu^3}{c^2}$$

thermodynamical equilibrium

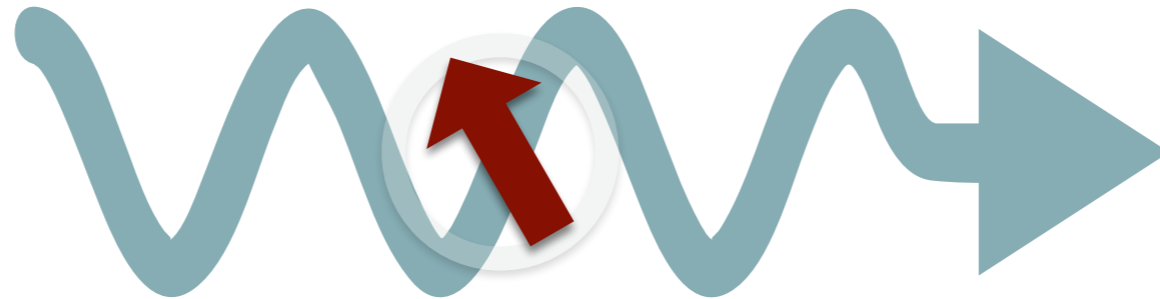
$$\frac{n_2}{n_1} = \frac{g_2}{g_1} \exp\left(-\frac{\epsilon}{\tau}\right)$$

$$\frac{n_1}{n_2} = \frac{g_1}{g_2} \exp\left(\frac{\epsilon}{\tau}\right)$$

$$p \propto \exp\left(-\frac{\epsilon_s}{\tau}\right)$$

$$A_{21} = \frac{64\pi^4\nu^3}{3hc^3} \mu^2 \frac{J+1}{2J+3}$$

**electric dipole moment  
can be calculated from wavefunction**



**radiation can only interact with charge**

(putting aside magnetic dipole moment)

# Line formation

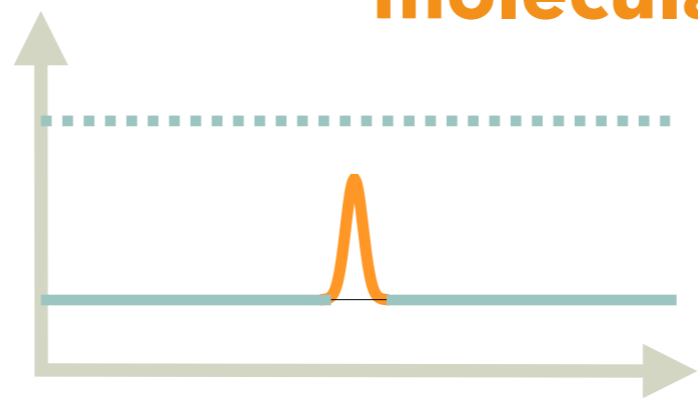
- 1** optical depth:  
when thick, we are always looking at the surface of  $\tau \sim 1$
- 2** warmer blackbody has stronger radiation than cooler one over all frequency
- 3** when the line intensity reached to that blackbody at the frequency, cannot grow further.

**Blackbody 11K > CO J=1-0 line at 10 K**

$T_k$

single temperature

molecular line  $\tau \sim 0.1$

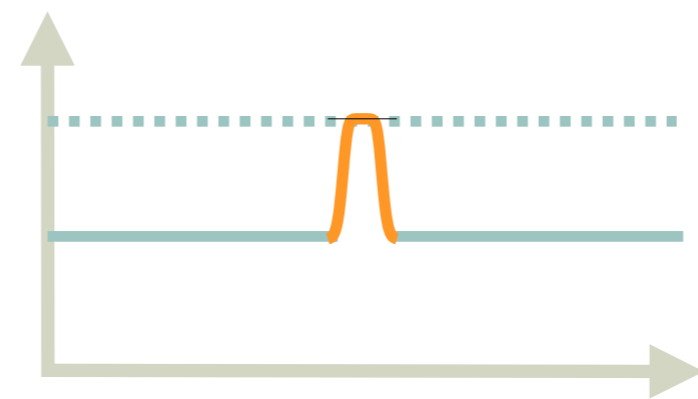


$T_k$  blackbody

gray body

$\tau \cdot B(T)$

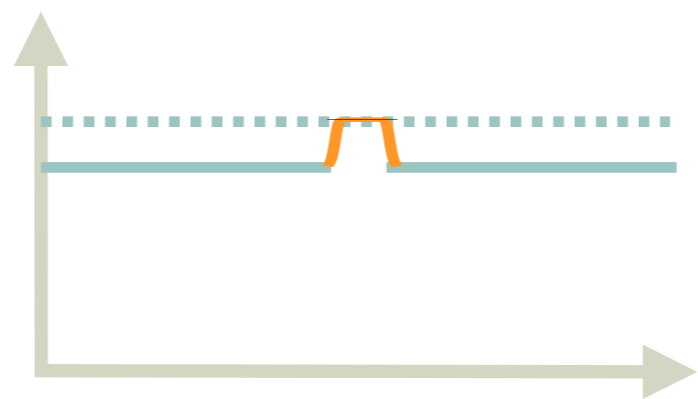
$\tau \ll 1$



$T_k$

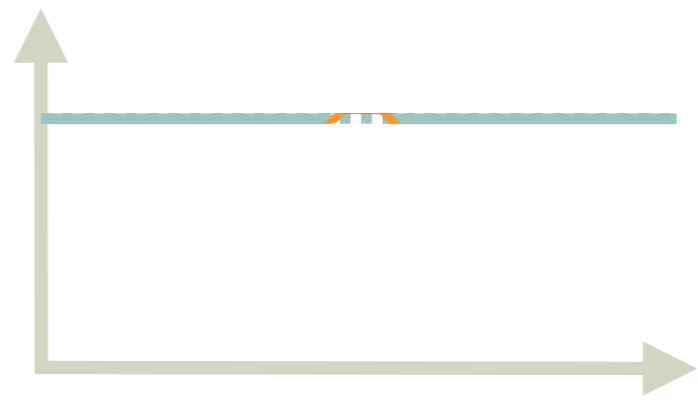
$\tau \sim 1$

$\tau \sim 0.1$



$\tau > 1$

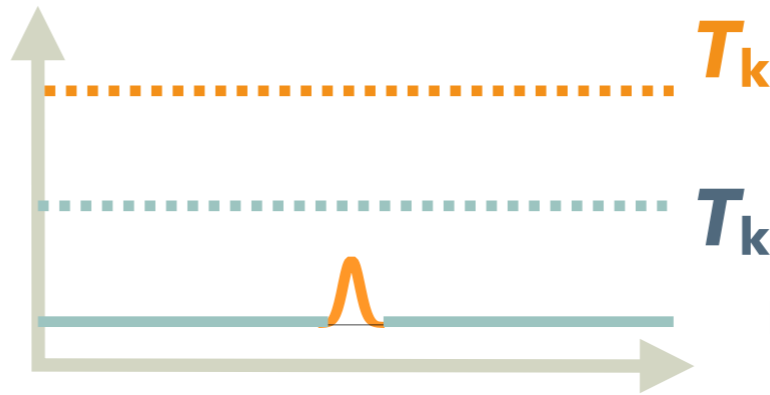
$\tau \sim 1$



$\tau > 1$

$\tau > 1$

temperature  
gradient

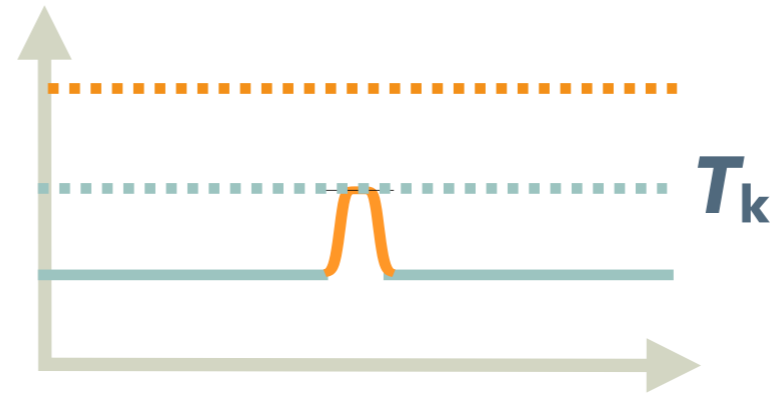


$T_k > T_k$   
blackbody

mixed gray body

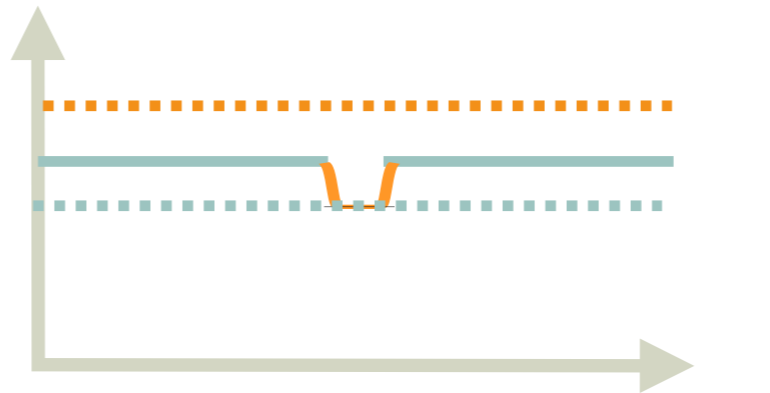
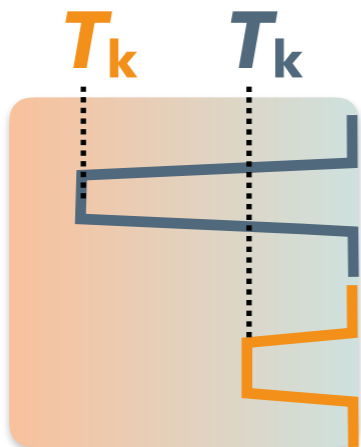
$$\tau \cdot B(T)$$

$$\tau \ll 1$$



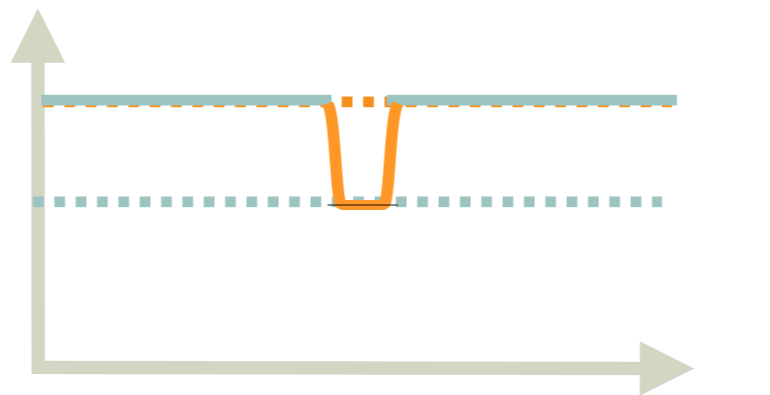
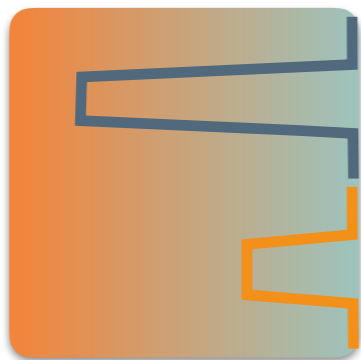
$$\tau \sim 1$$

$$\tau \sim 0.1$$

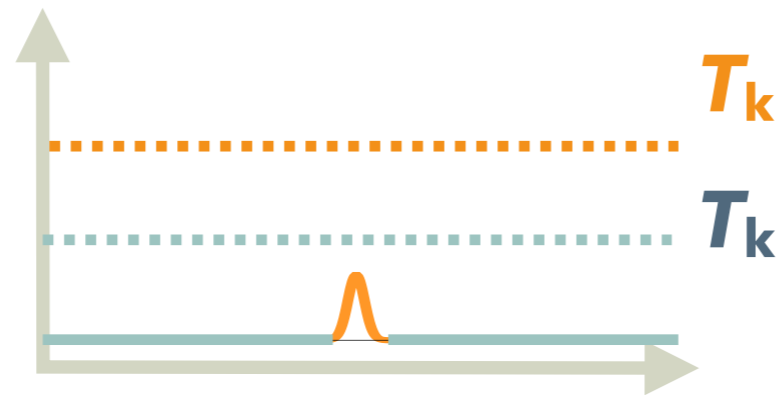
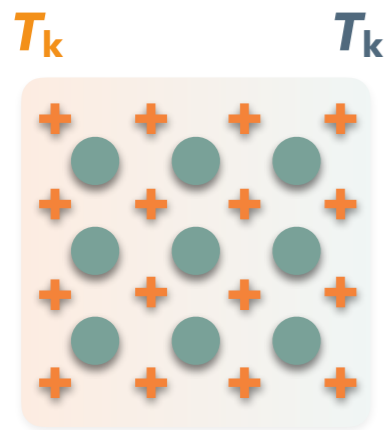


$$\tau > 1$$

$$\tau \sim 1$$

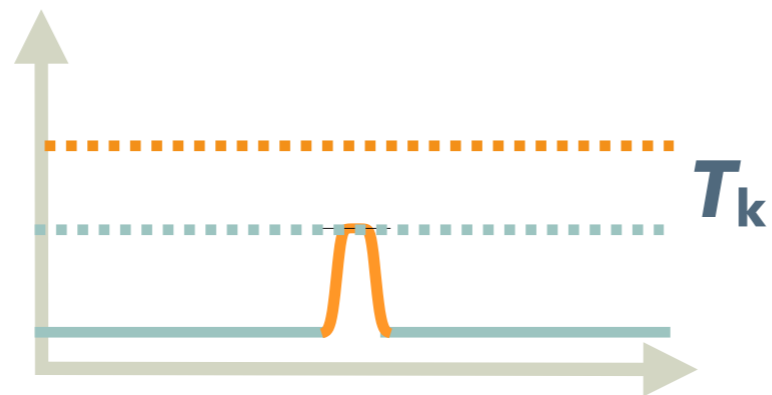
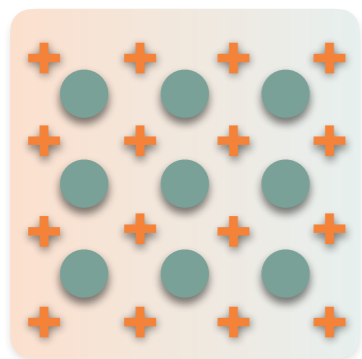


$$\tau > \tau > 1$$

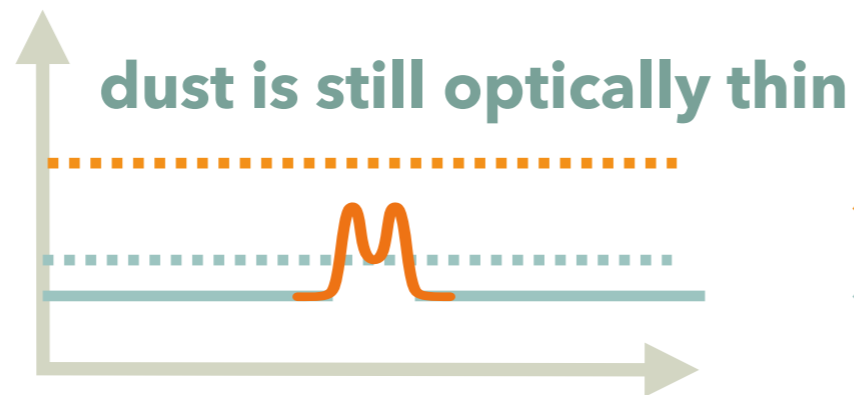
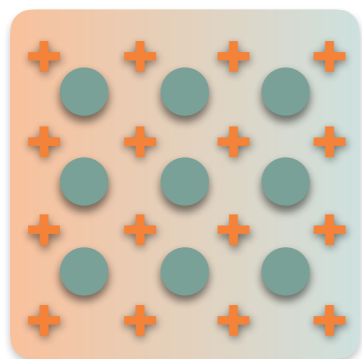


● dust (blackbody)  
+ molecules (line)

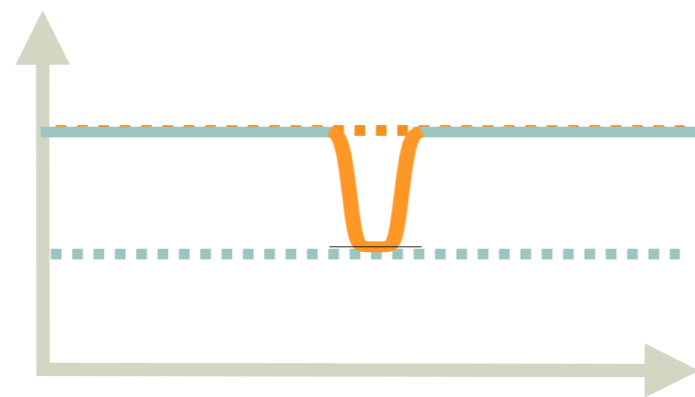
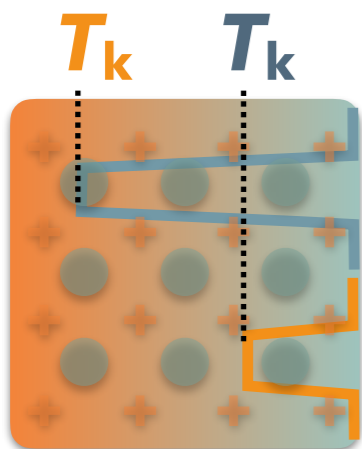
$\tau \cdot B(T)$   
 $\tau \ll 1$



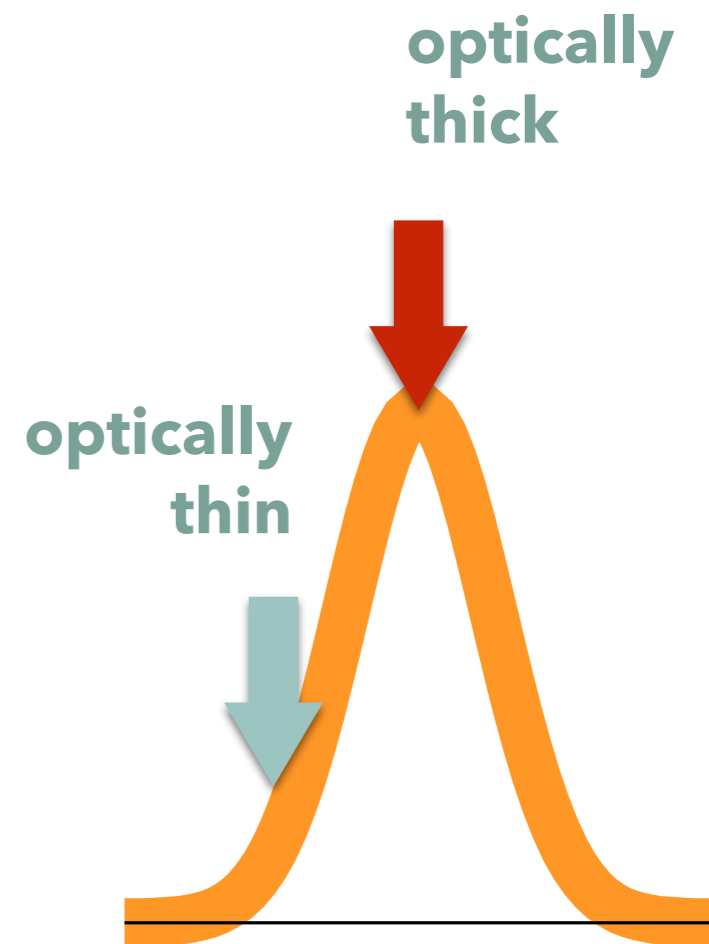
$\tau \sim 1$   
 $\tau \sim 0.1$



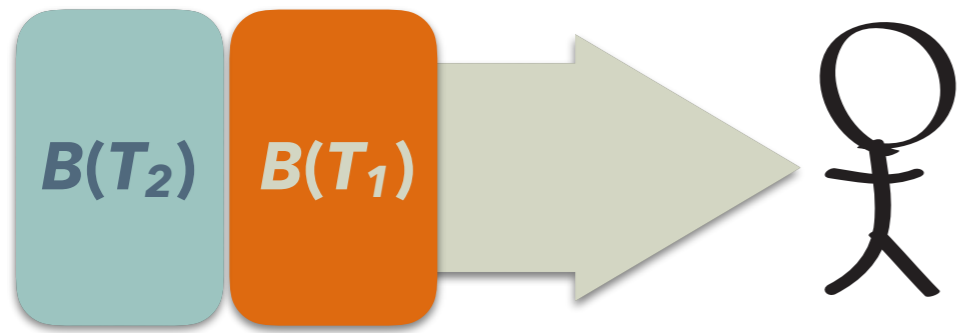
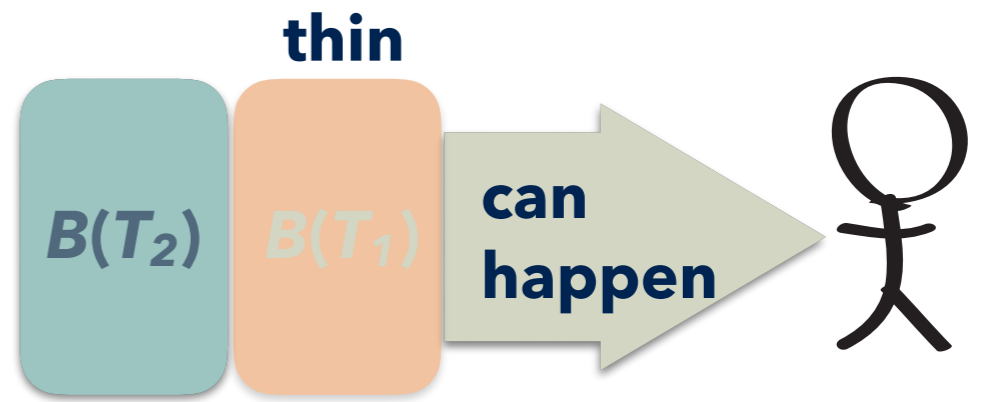
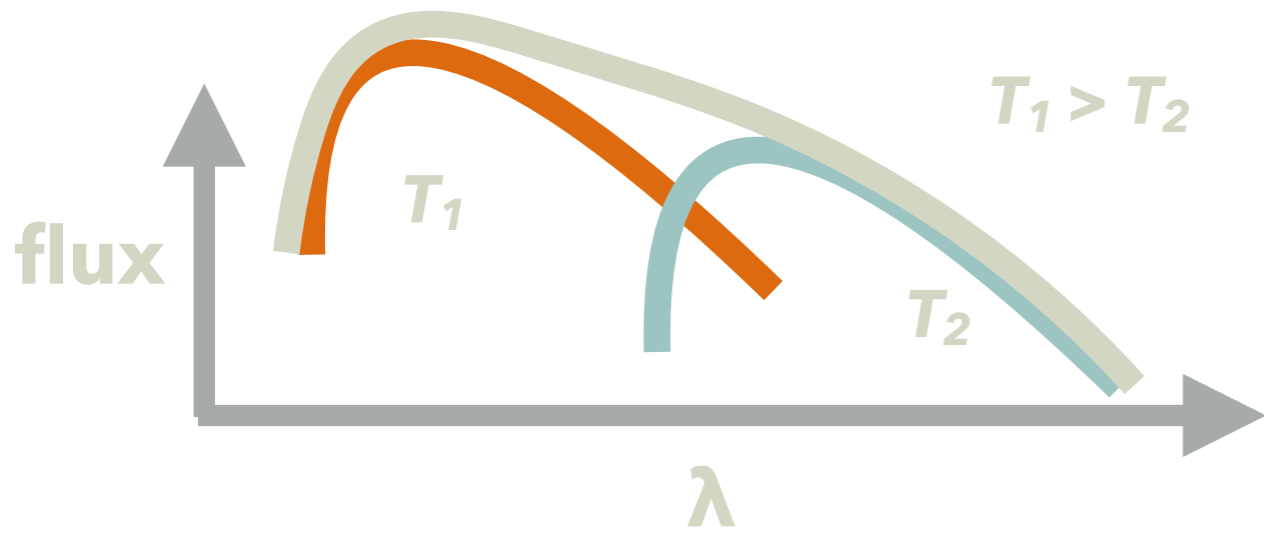
$\tau > 1$   
 $\tau \sim 1$



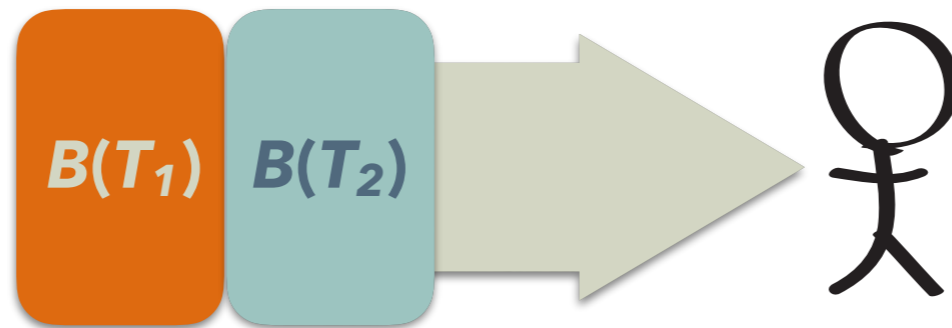
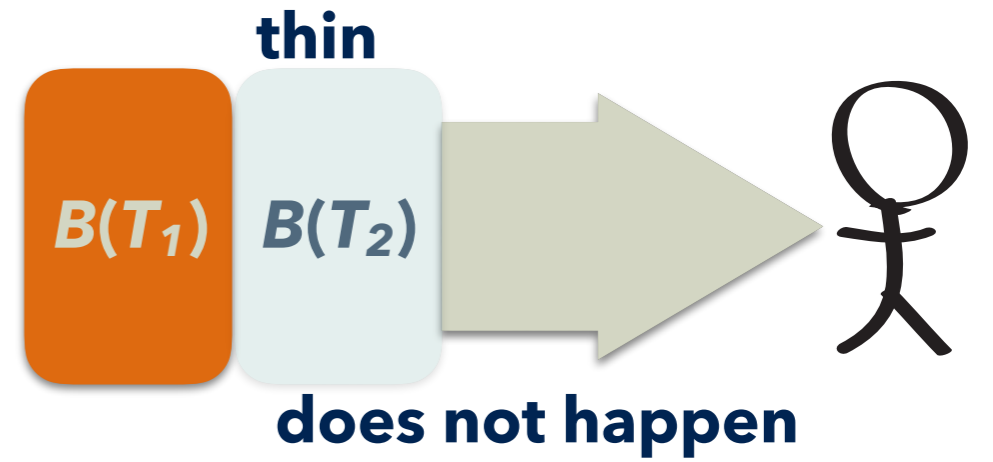
$\tau > 1$   
 $\tau > 1$





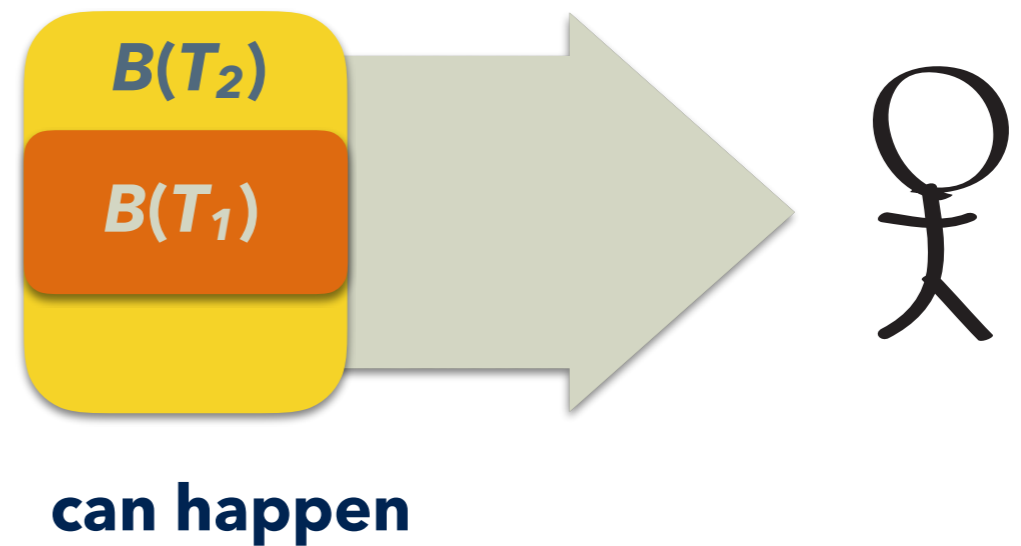
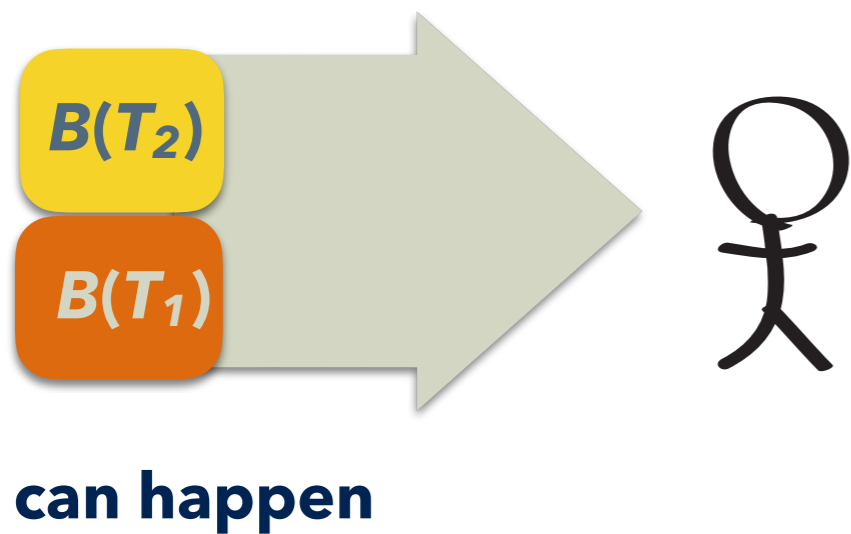
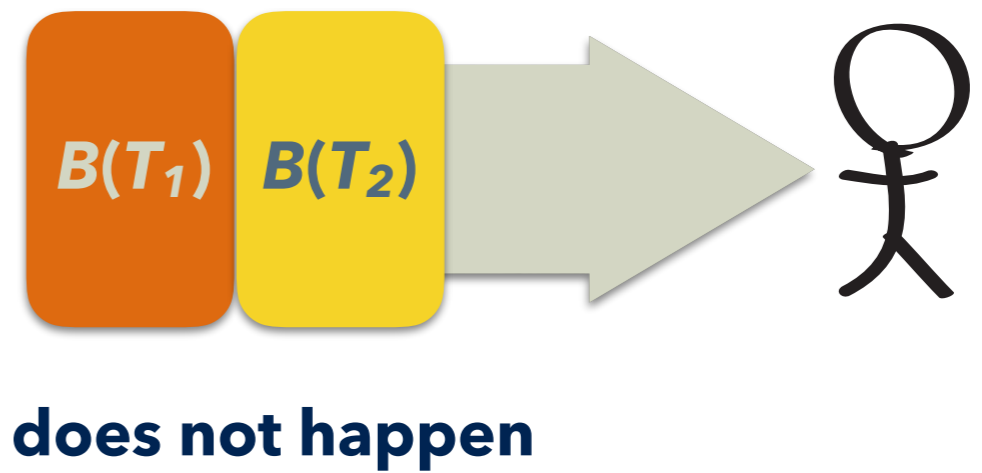
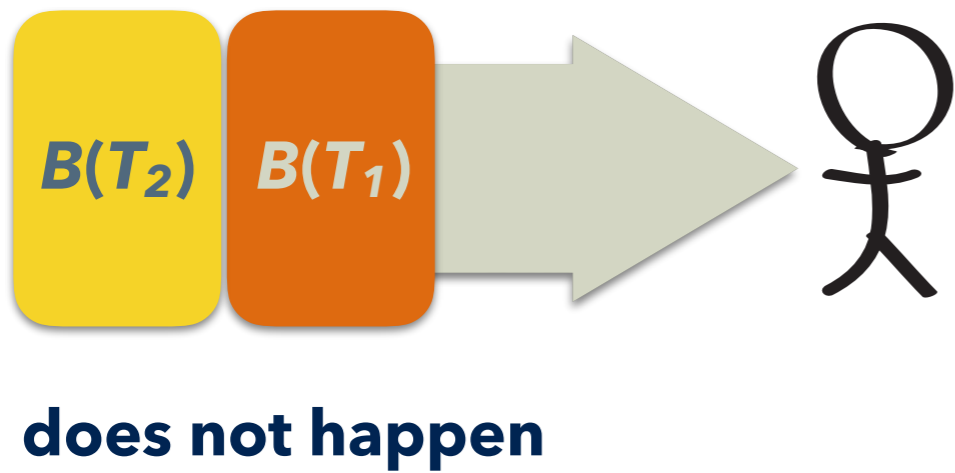
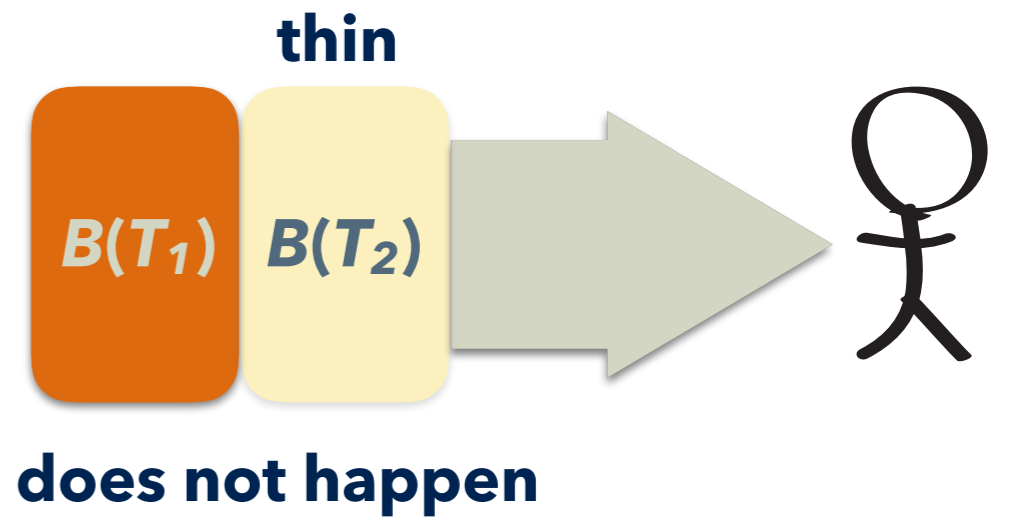
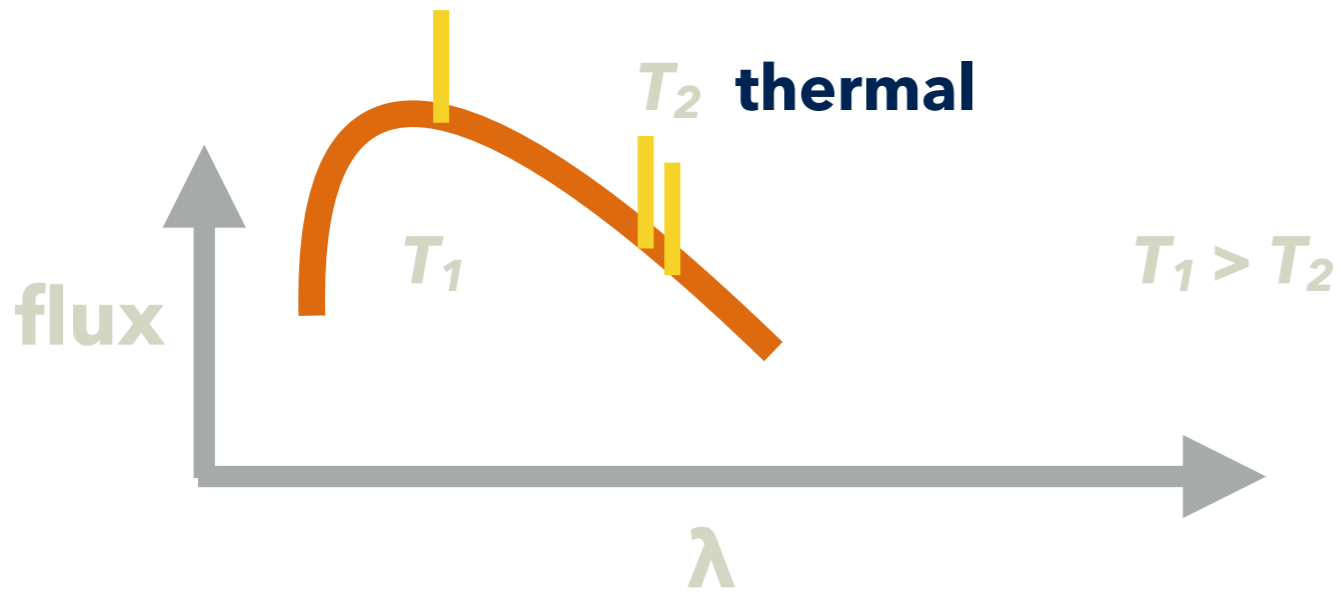


**does not happen**



**does not happen**





# Exercise today

## 1 Stefan-Boltzmann constant

$$\sigma = \frac{2\pi^5 k^4}{15c^2 h^3}$$

$$\int_0^\infty \frac{x^3}{e^x - 1} dx = \frac{\pi^4}{15}$$

Planck function

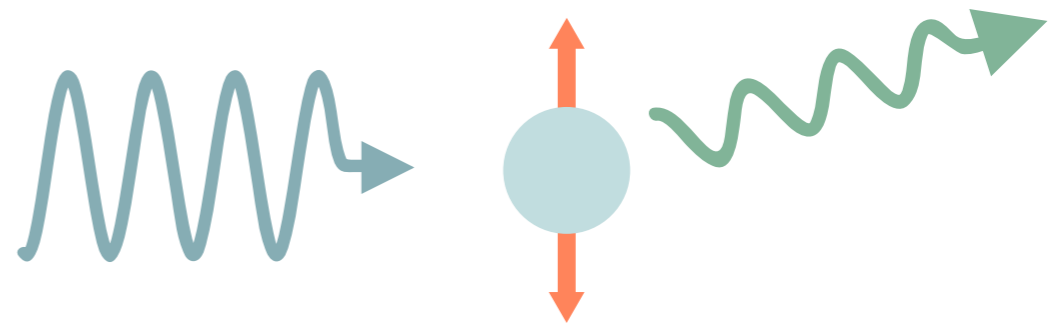
$$B(\nu) = \frac{2h\nu^3}{c^2} \frac{1}{\exp \frac{h\nu}{kT} - 1}$$

## 2 Why Thomson scattering is sometimes referred to "electron" scattering?

Thomson cross section

$$P = \langle S \rangle \sigma_T$$

$$\sigma_T = \frac{P}{\langle S \rangle} = \frac{8\pi}{3} \frac{e^4}{m^2 c^4}$$



we did not  
assume it is an  
electron,  
but a charge

$$P = \frac{e^4 E_0^2}{3m^2 c^3}$$

## Dipole approximation

$$\langle \sin^2 \omega t \rangle = \frac{1}{2}$$

$$\frac{dP}{d\Omega} = \frac{\ddot{\mathbf{d}}^2}{4\pi c^3} \sin^2 \Theta$$

## incoming electric field

$$\mathbf{E} = \boldsymbol{\varepsilon} E_0 \sin \omega t$$

$$m\ddot{\mathbf{r}} = e\mathbf{E}$$



## dipole moment

$$\mathbf{d} = e\mathbf{r}$$

$$\ddot{\mathbf{d}} = e\ddot{\mathbf{r}}$$

$$\ddot{\mathbf{d}} = \frac{e^2 \mathbf{E}}{m}$$

$$|\ddot{\mathbf{d}}|^2 = \frac{e^4 E_0^2}{2m^2}$$

**3** Find

$$\frac{dl}{d\tau} = -I + S$$

$$I = I(0) e^{-\tau} + S(1 - e^{-\tau})$$

Set  $\mathcal{I}_\tau = I_\tau e^\tau$

$$S = S e^\tau$$

note  $S$  is not dependent on  $\tau$

$$\frac{d\mathcal{I}_\tau}{d\tau} = \frac{dI_\tau}{d\tau} e^\tau + I_\tau e^\tau$$

$$\frac{dS}{d\tau} = S e^\tau$$

